

Multiresponsive Behavior of Membranes and Vesicles

Reinhard Lipowsky

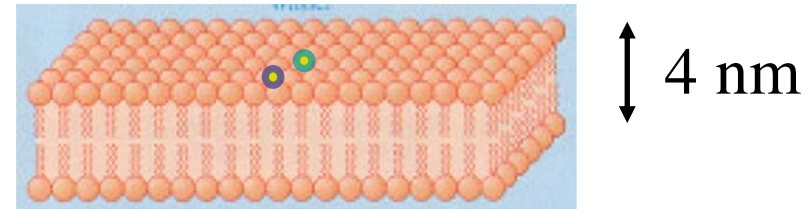
MPI of Colloids and Interfaces, Potsdam-Golm

- Intro: Fluid Membranes and GUVs
- GUVs and Aqueous Two-Phase Systems
- Spontaneous Tubulation and Tension
- Morphological Complexity
- Wetting and Fluid-Elastic Scaffolding
- GUVs and Biomolecular Condensates
- Endocytosis of Nanodroplets
- Endocytosis of Nanoparticles
- Outlook: Droplet-Stabilized GUVs

Biomembranes are Fluid Bilayers

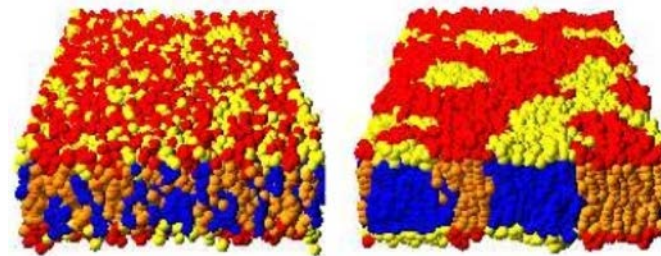
- **Fluid** membranes, i.e.,
fast lateral diffusion:

Diffusion constant $\sim \mu\text{m}^2/\text{s}$

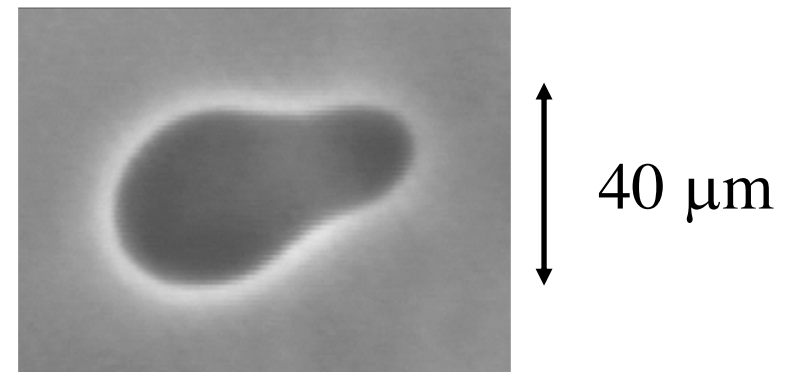


lipid swapping $\sim \text{ns}$

- Lateral diffusion =>
Compositional responses,
demixing, domain formation ...

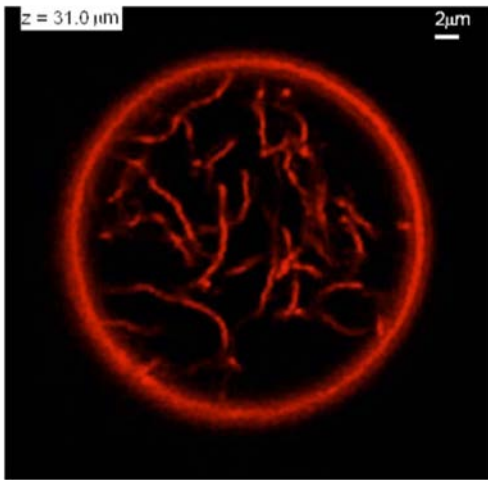


- Flexibility =>
Morphological responses,
budding, tubulation, ...
Direct evidence for fluidity

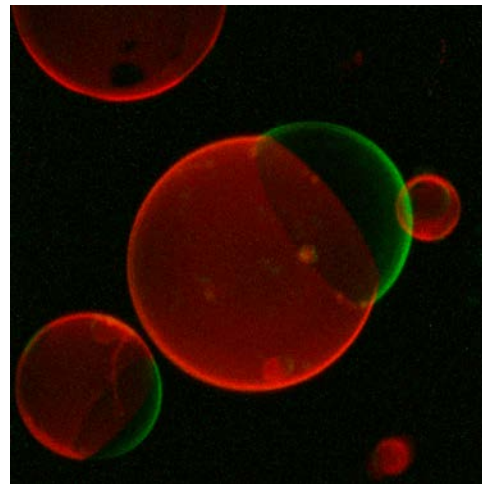


Multiresponsive Behavior of GUVs

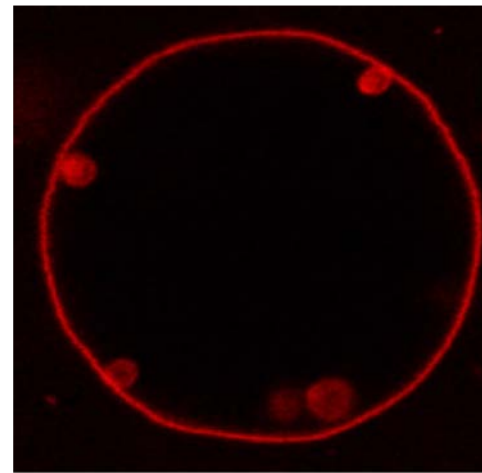
- Giant unilamellar vesicles (GUVs), tens of micrometers
- Remodelling in response to various perturbations:



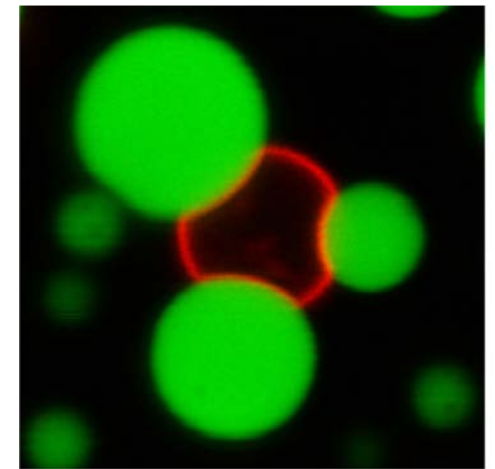
Nanotubes from polymer adsorption, tube width ~ 100 nm



Formation of intra-membrane domains, 2D phase separation



Small buds from adsorption of two ESCRT proteins

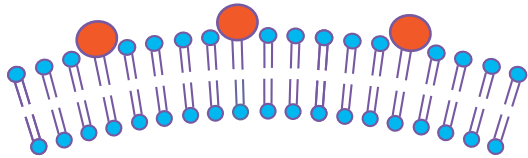


Shaping GUVs by biomolecular condensates

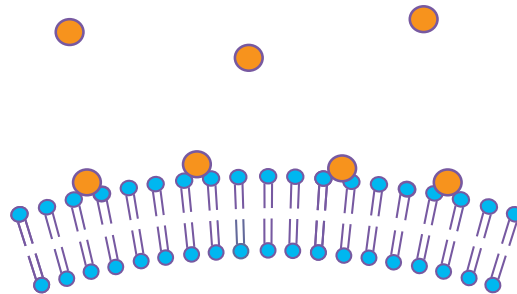
- What are the forces that drive remodelling processes?

Spontaneous = Preferred Curvature

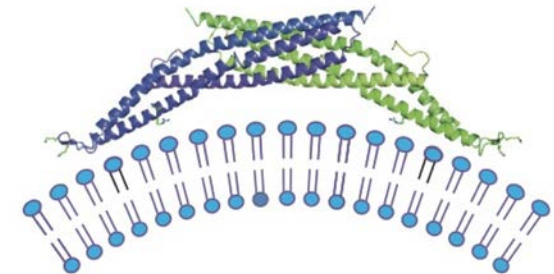
- Spontaneous or preferred curvature m describes bilayer asymmetry = asymmetry between two leaflets
- Different molecular mechanisms for bilayer asymmetry:



Asymmetric
composition,
e.g., ganglioside



Asymmetric
adsorption of
small molecules



Asymmetric
protein coats,
e.g. BAR-domain

Curvature Elasticity

Helfrich, Z. Naturforschung (1973)

- Local mean curvature M tries to adapt to spontaneous (or preferred) curvature m
- Curvature or bending energy:

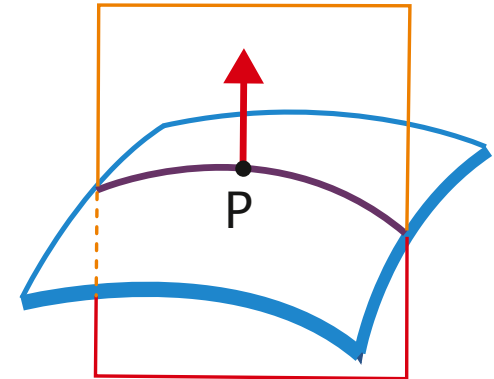
$$E_{cu} = \int dA 2 \kappa (M - m)^2$$

integral over membrane area A

- 2nd fluid-elastic parameter: Bending rigidity κ
Dimensions of energy, $\kappa = 10^{-19} \text{ J} = 20 k_B \text{ T}$

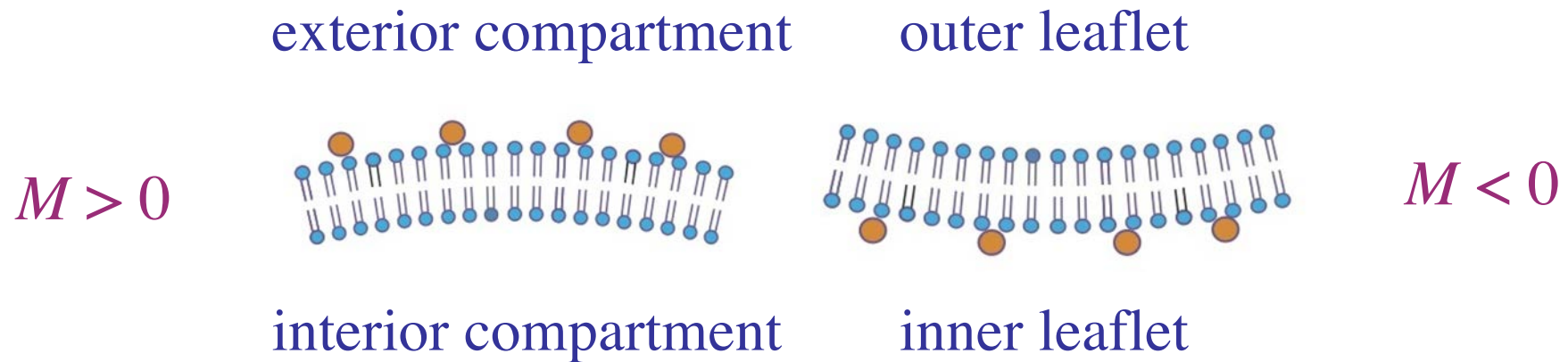
- Range of spontaneous curvatures m

from $1/(20 \text{ nm})$ to $1/(20 \mu\text{m})$



Sign of (Spontaneous) Curvature

- Mean curvature M and spontaneous curvature m can be positive or negative
- Sign defined with respect to interior/exterior compartments = with respect to inner/outer leaflet



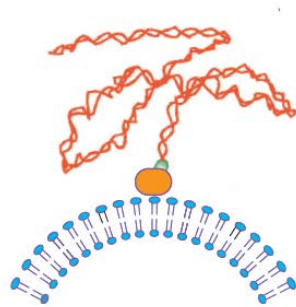
Mean curvature M is positive (negative) if membrane bulges towards exterior (interior) compartment

Local Curvature Generation

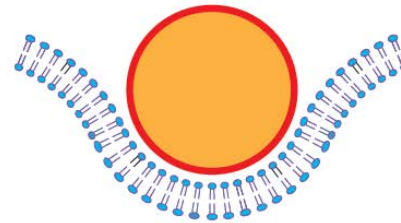
RL, *Faraday Disc.* (2013); *Biol. Chem.* (2014)

- Local curvature generated on nanoscopic scales:

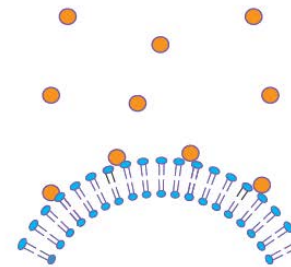
Anchored polymer



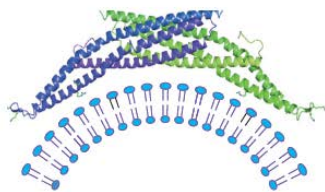
'Large' particle



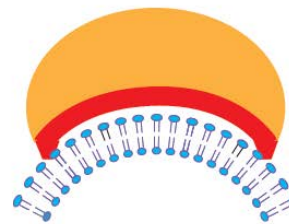
'Small' adsorbate particles



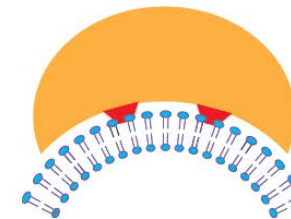
BAR-domain protein



Nonspherical Janus particles



Induced Fit



Conformational Selection

Spont Curv from Mol Simulations

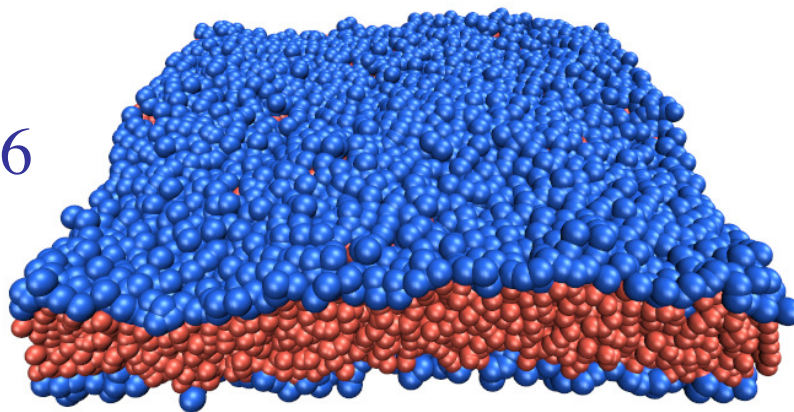
Rozycki, RL, *J. Chem. Phys.* (2015); *J. Chem. Phys.* (2016)

- Stress profiles and tension-free states
- Spont curv from first moment of stress profiles
- Example: Different leaflet densities

Asymmetry $\phi = N_{\text{ex}} / (N_{\text{ex}} + N_{\text{in}})$

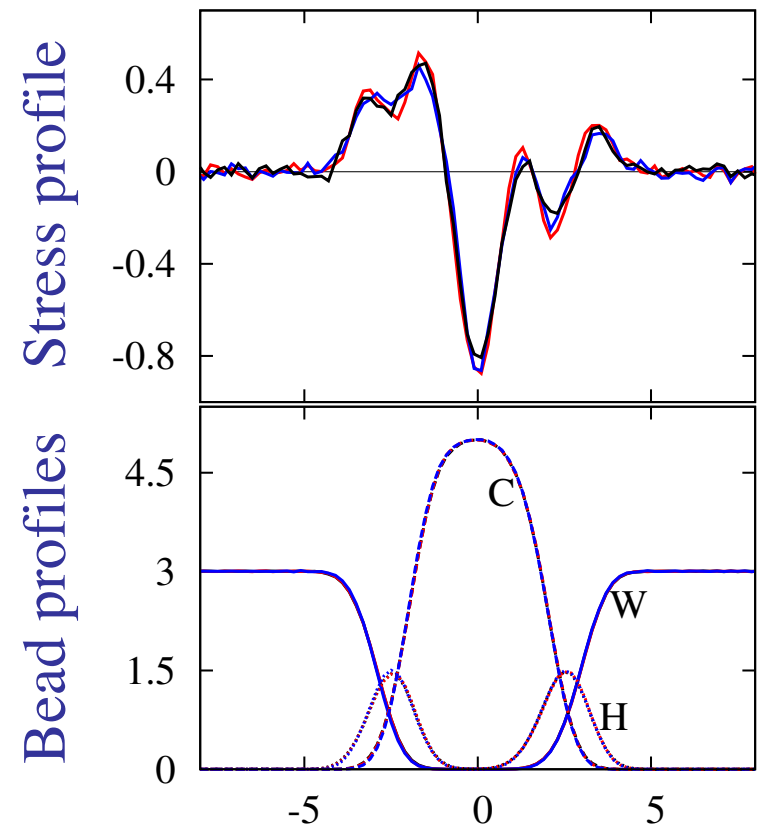
outer leaflet with N_{ex} lipids

$\phi = 0.56$

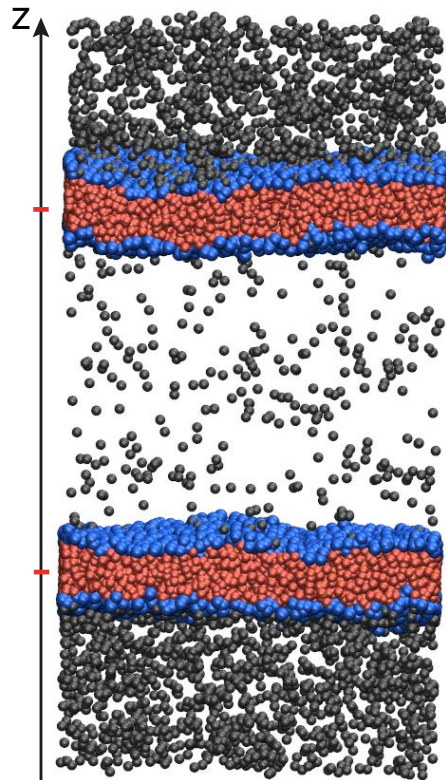


inner leaflet with N_{in} lipids

$\phi = 0.51$



Asymmetric Adsorption and Depletion



Particle concentration X_{ex}

Bilayer 1

Particle concentration X_{in}

Bilayer 2

Particle concentration X_{ex}

- Spont curv proportional to $\pm (X_{\text{ex}} - X_{\text{in}}) = \pm \Delta X$

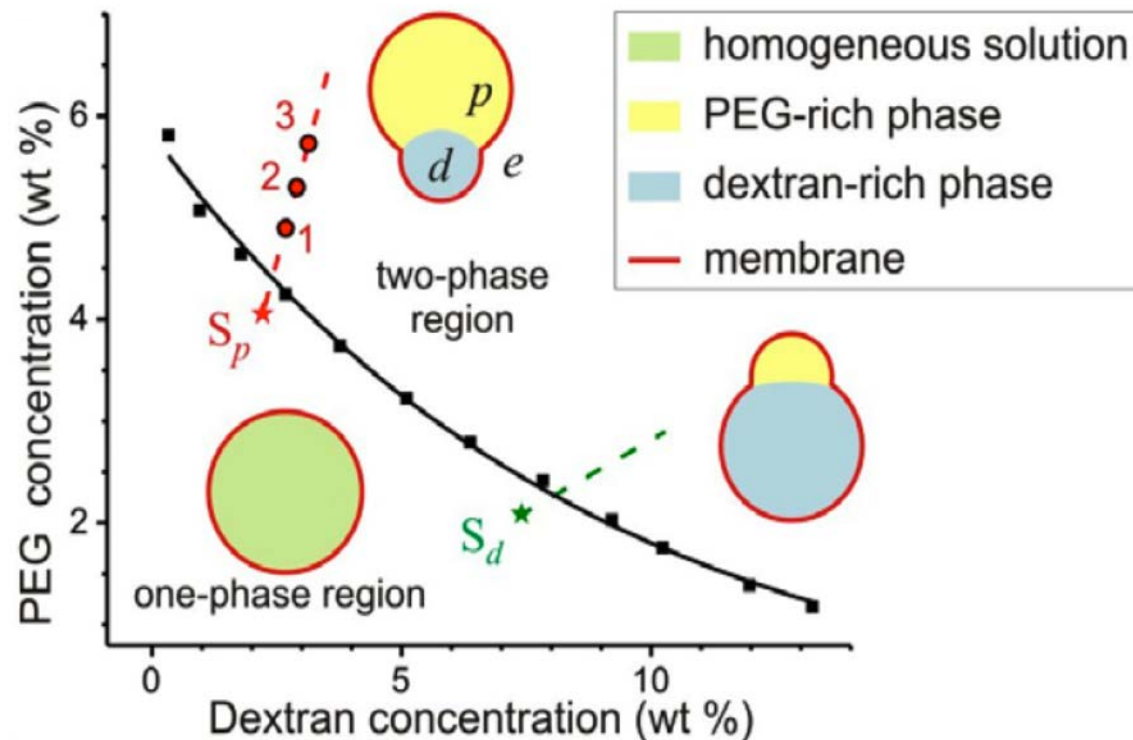
- Example: 1 nm particles, $\Delta X = 100 \text{ mM}$

Adsorption: $m = 1/(77 \text{ nm})$, Depletion: $m = -1/(270 \text{ nm})$

GUVs + Aqueous Phase Separation

Li, RL, Dimova, *JACS* (2008); *PNAS* (2011)
Liu, Agudo. ... RL, *ACS Nano* (2015)

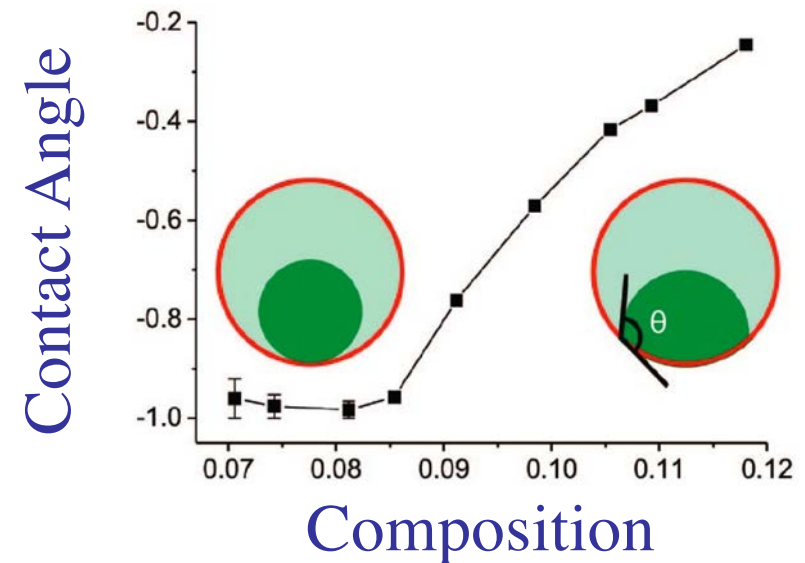
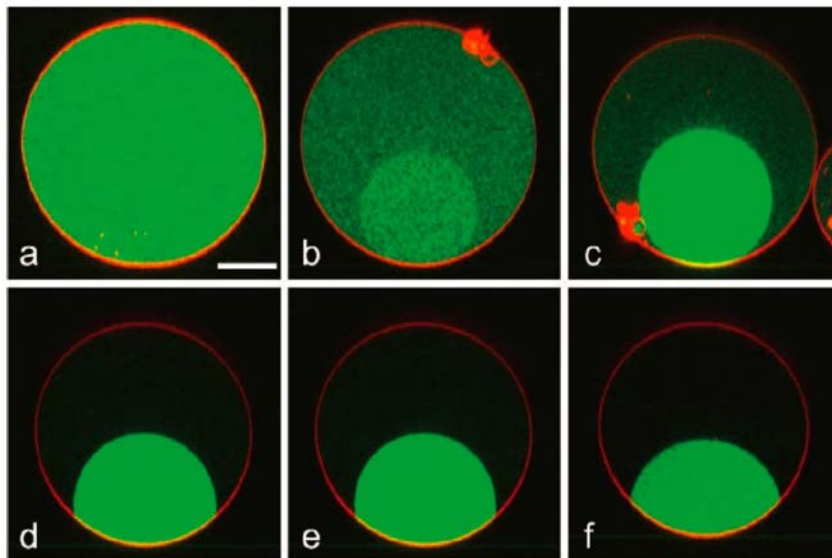
- GUVs filled with aqueous polymer solution
- Example: PEG and dextran
- Increase polymer concentration via deflation:



1st Surprise: Wetting Transition

Li et al, *JACS* (2008)

- Shape evolution for vesicle during deflation:

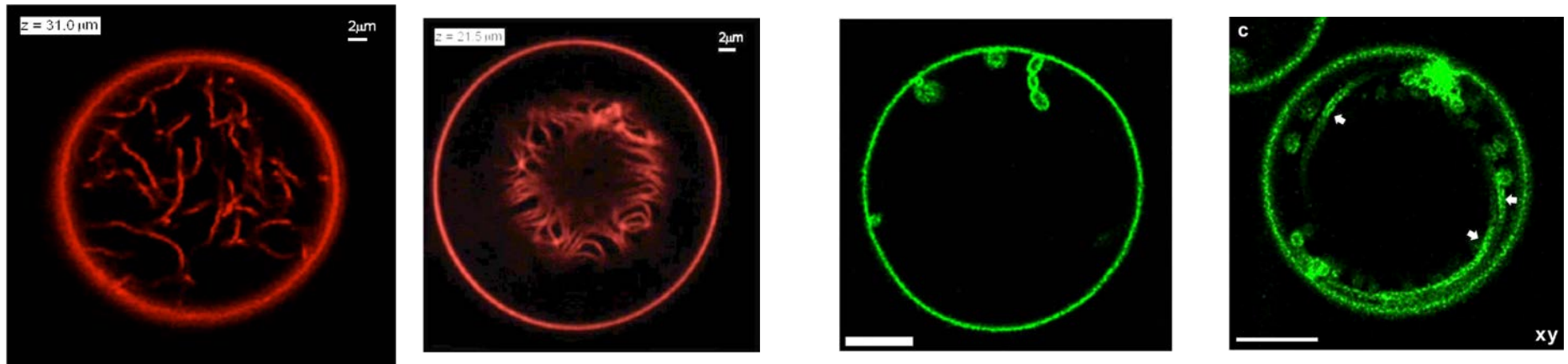


- Low deflation: zero contact angle \Rightarrow complete wetting
- High deflation: finite contact angle \Rightarrow partial wetting

2nd Surprise: Membrane Nanotubes

Li et al, *PNAS* (2011) Liu et al, *ACS Nano* (2016)

- Membranes labeled by fluorescent dyes
- Lipid mixture of DOPC, DPPC, cholesterol
- Liquid-disordered (red) and liquid-ordered phase (green)

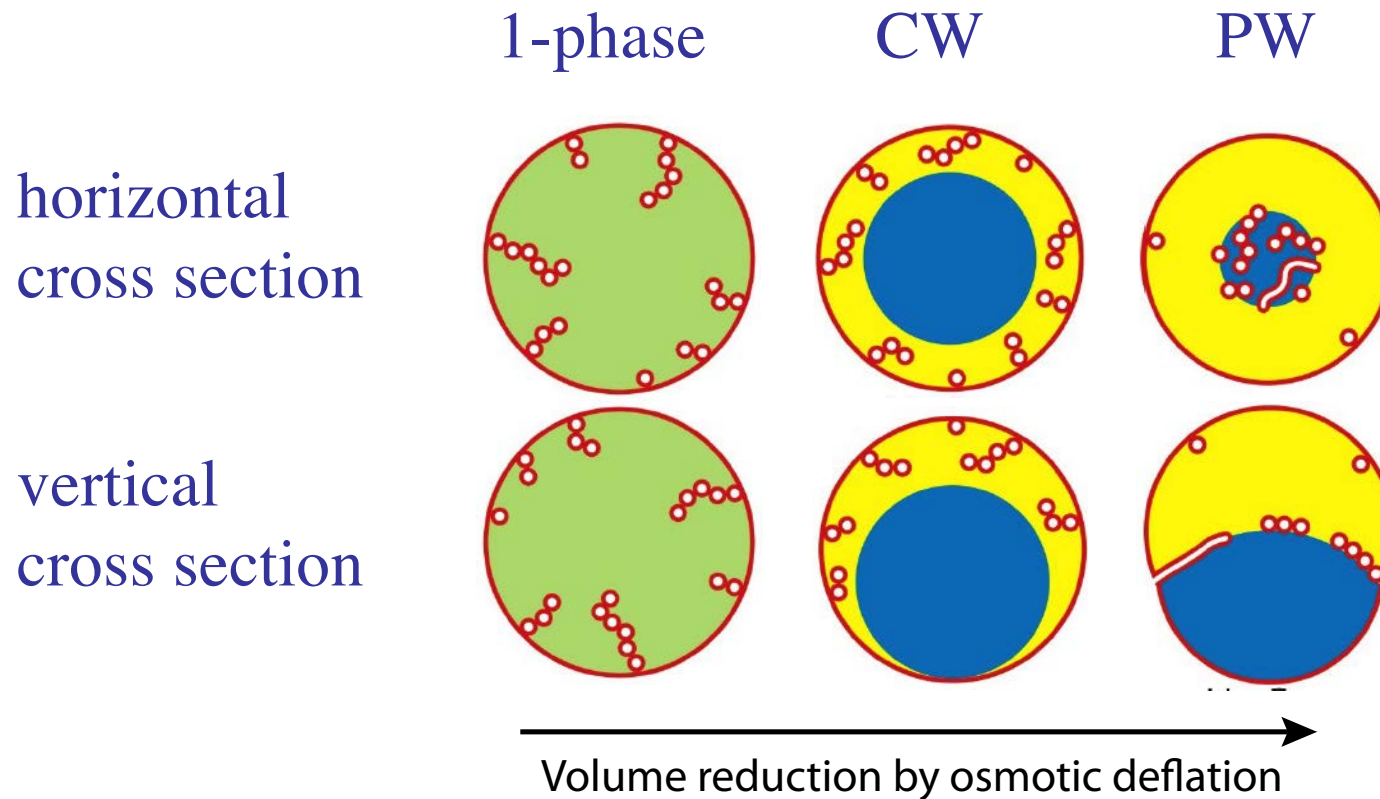


- Spontaneous tube formation **without** external forces
- Inward-pointing tubes reveal large negative spont curv
- Tubes can be necklace-like or cylindrical

Tubulation and Wetting

Liu, Agudo ... RL, *ACS Nano* (2015)

- Nanotubes form for different wetting morphologies:



- Tubulation does **not** require aqueous phase separation

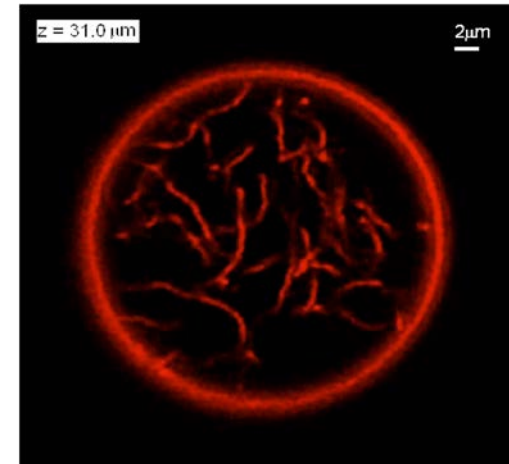
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Spont Tubulation and Tension

RL, *Faraday Discuss.* (2013)

- Tubulation leads to tense mother vesicle
- Total tension in Euler-Lagrange equation has two components:

$$\hat{\Sigma} = \Sigma + \sigma$$



Mechanical tension Σ stretches the membrane

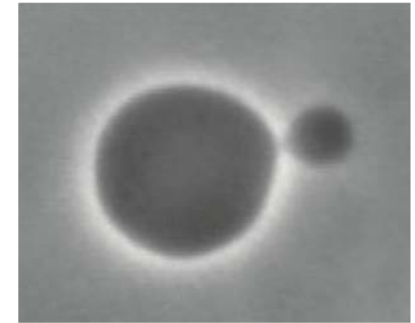
Spontaneous tension $\sigma = 2 \kappa m^2$ for $M \ll m$

- Presence of nanotubes implies dominance of spontaneous tension, mechanical tension can be ignored
- Example: Spont curvature $\approx -1/(100 \text{ nm})$
 - Spontaneous tension $\sigma \approx 10^{-2} \text{ mN/m}$
 - Mechanical tension $\Sigma \approx 10^{-4} \text{ mN/m}$

How Do Nanotubes Form?

Liu et al, *ACS Nano* (2016)

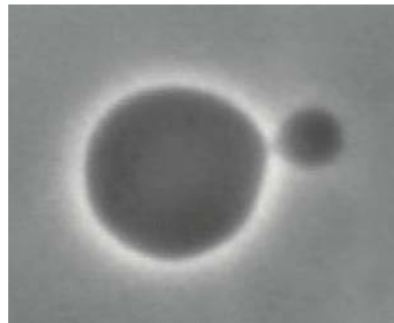
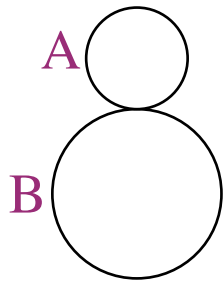
- Tubulation intimately related to budding
- Osmotic deflation of spherical GUV
- Small deflation step leads to single bud
- Bud and mother vesicle connected by membrane neck
- Bud acts as nucleation site for necklace-like tube
- Several pathways for subsequent deflation steps:
 - Formation of new bud
 - Bud into 2-necklace
 - N-necklace into (N+1)-necklace



Membrane Buds and Necks

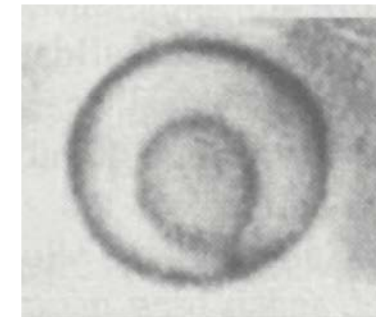
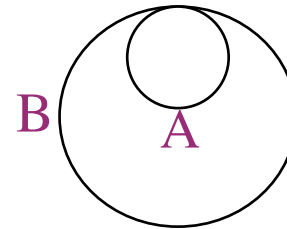
- For $m \neq 0$, curvature elasticity leads to spherical membrane segments connected by membrane necks

- Out-bud:



spont curv $m > \sqrt{2}/R_{ve}$

- In-bud:



spont curv $m < 0$

- Closed neck is stable if:

$$0 < M^A + M^B \leq 2m$$

$$2m \leq M^A + M^B < 0$$

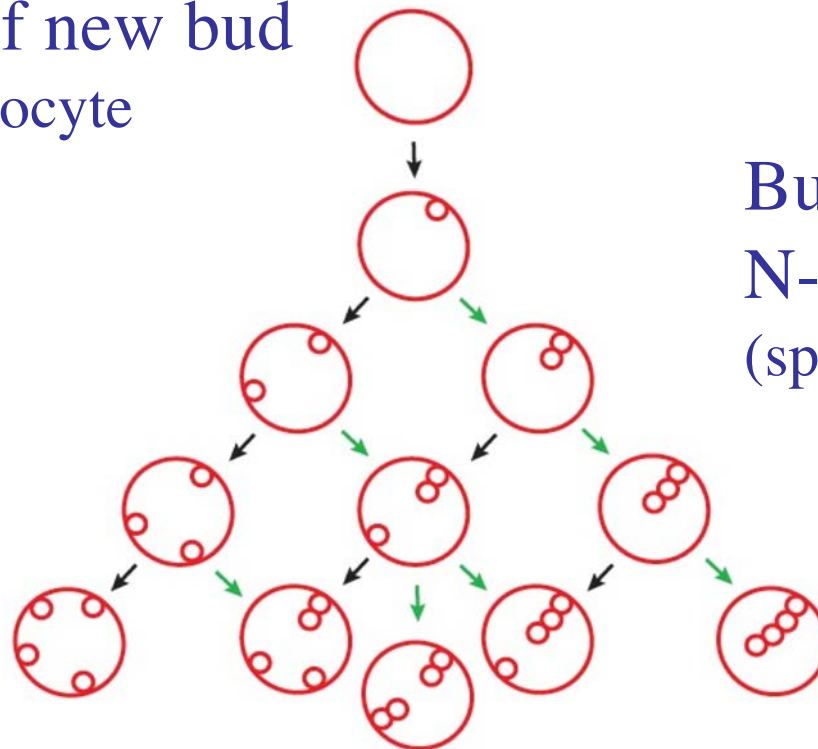
- Simple relation between geometry and material parameter

Nucleation and Growth of Tubes

Liu et al, *ACS Nano* (2016)
RL, *J. Phys. D* (in press)

- Spherical GUV, large spont curv m
- Osmotic deflation of GUV in discrete steps
- At each step, different morphological pathways:

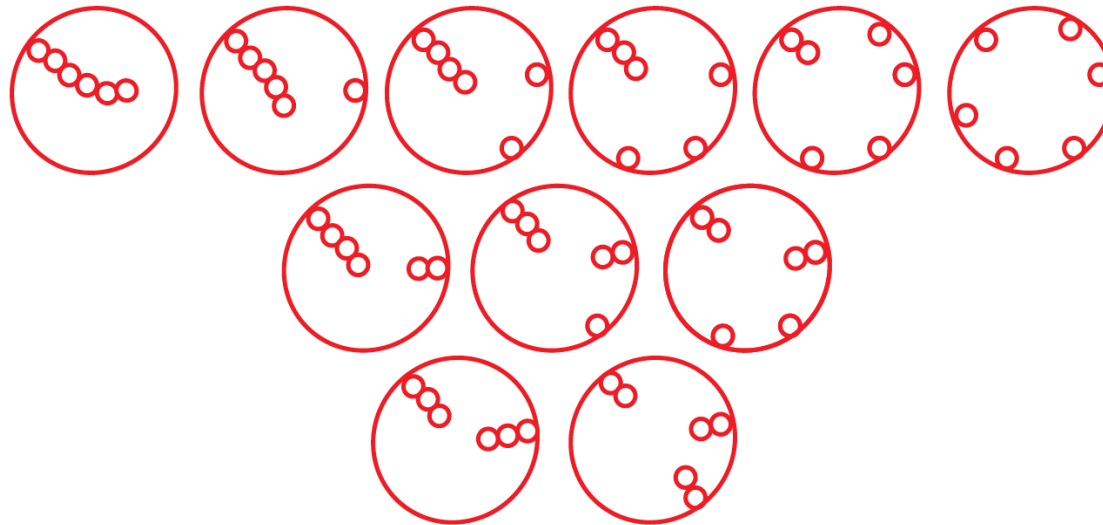
Formation of new bud
(oblate-stomatocyte
bifurcation)



Bud into 2-necklace
N- into (N+1)-necklace
(sphere-prolate bifurcation)

Morphological Complexity: Theory

- After 6th step, 11 morphologies with 6 spherules:



- All beads are connected by membrane necks
- All morphologies have the same area, volume, and curvature energy
- Rugged energy landscape contains 11 intersecting branches
- For large N , number of N -spherule morph grows as $\exp[c \sqrt{N}]$

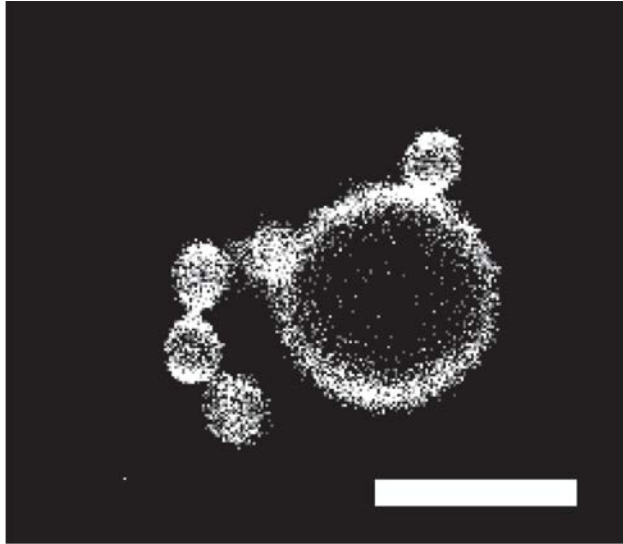
Morph Complexity: Experiment

Tripta Bhatia
(unpublished)

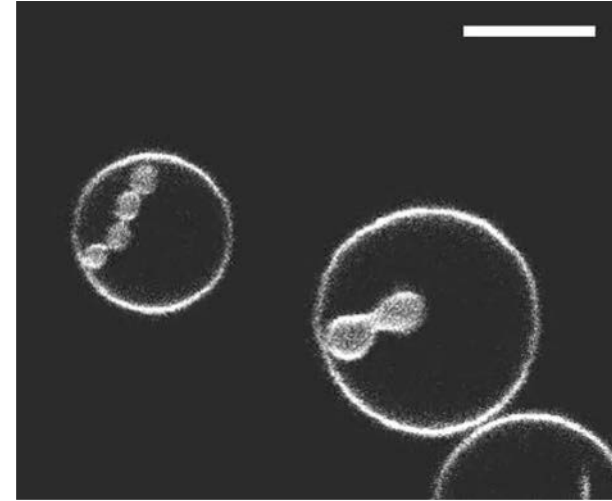
- Out-Necklaces

- In-Necklaces

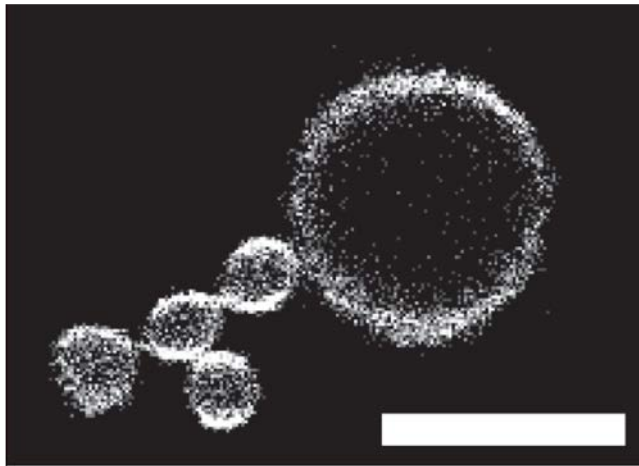
Linear



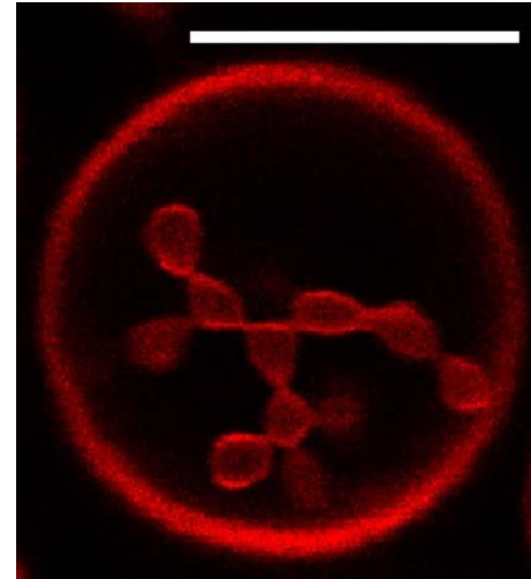
Linear



Branched

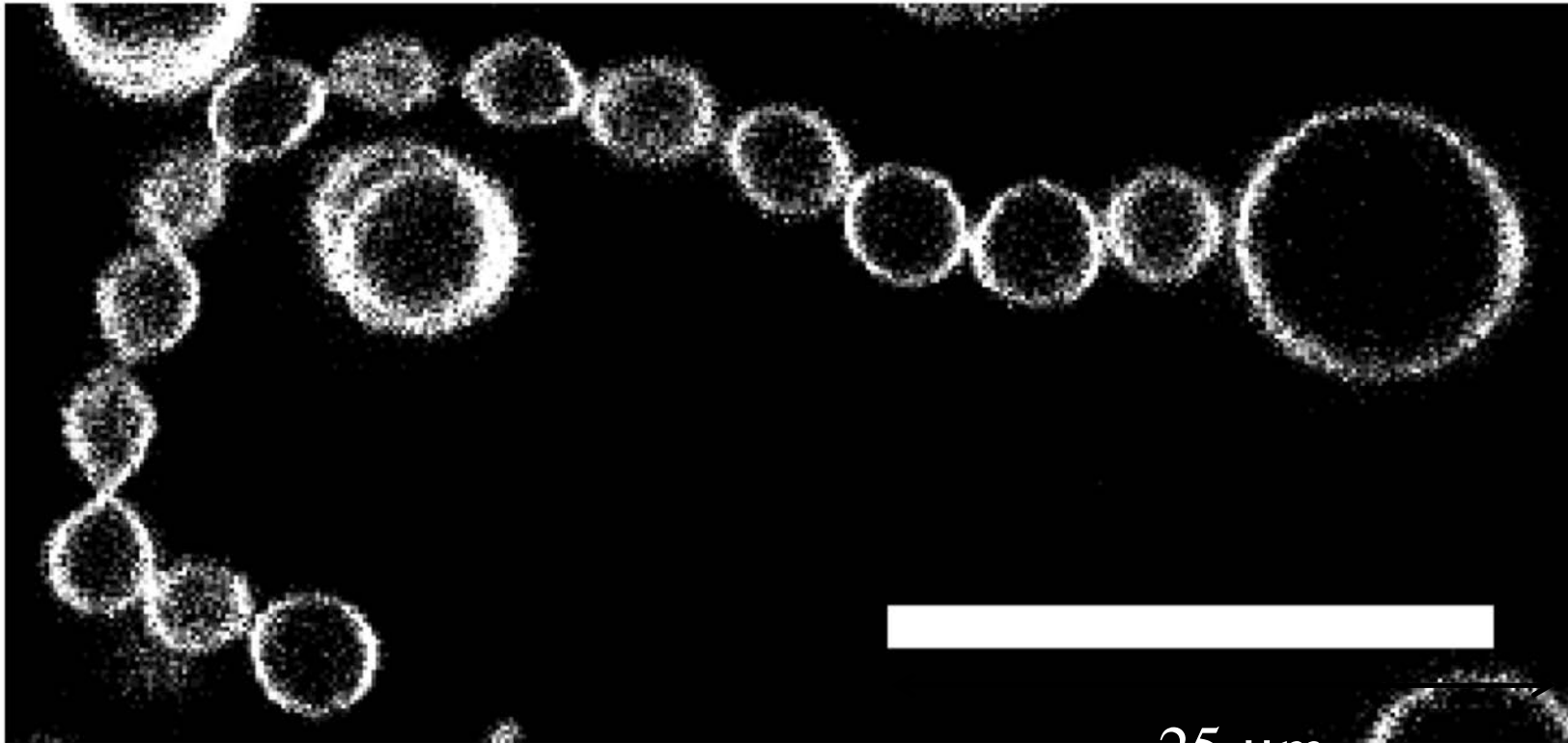


Branched



Long Necklaces

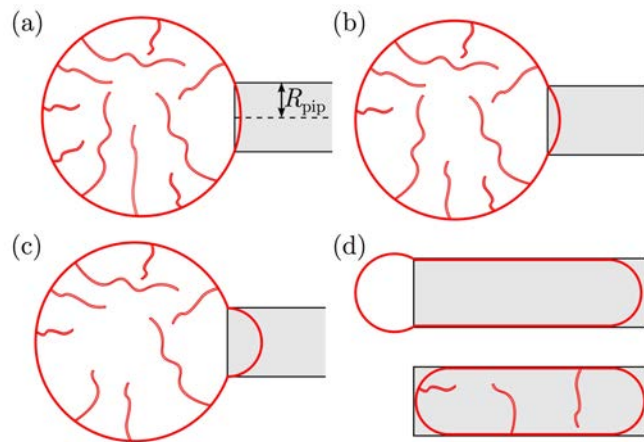
- Example: Out-Necklace with 14 beads



- Initial deflation leads to one bud
- Bud grows into long necklace

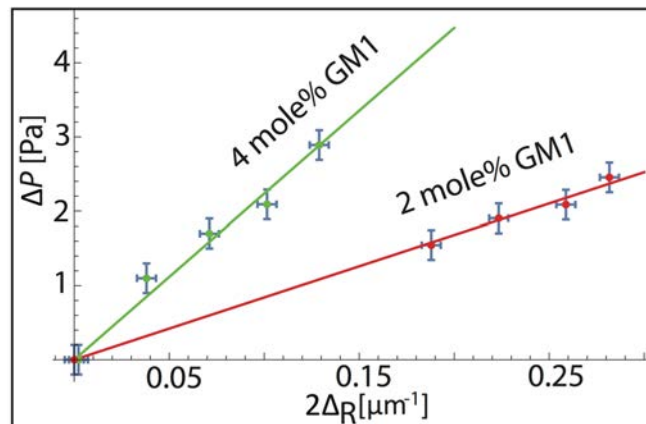
Nanotubes Increase Robustness of GUVs

- Retraction of tubes by micropipettes: Bhatia et al, *ACS Nano* (2018)



Initial aspiration up to hemispherical tongue then vesicle starts to flow into micropipette, increased robustness !

Suction pressure



Initial aspiration:

Aspiration pressure versus geometric quantity Δ_R

Slope = spontaneous tension σ

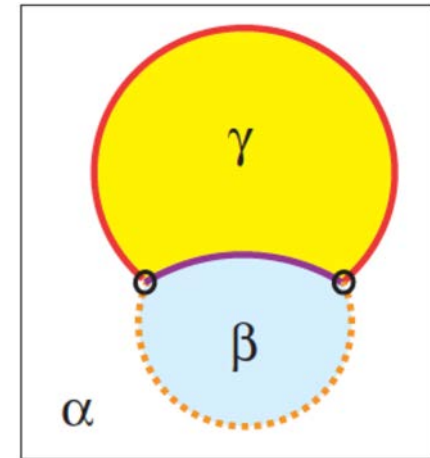
- Tubulated GUV behaves like a liquid droplet
- Interfacial tension = spontaneous tension

- Intro: Membranes and Giant Vesicles (GUVs)
- GUVs and Aqueous Two-Phase Systems
- Spontaneous Tubulation and Tension
- Morphological Complexity
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- Endocytosis of Nanoparticles
- Outlook: Droplet-Stabilized GUVs

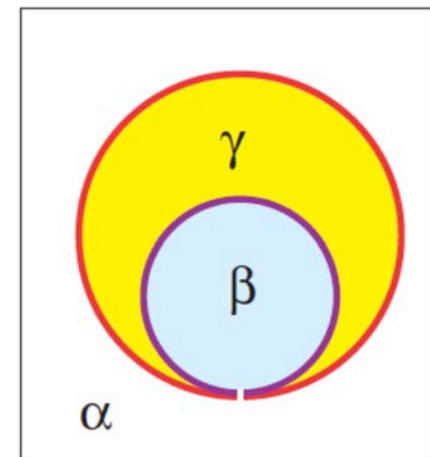
Membranes and Droplets

RL, *J. Chem. Phys. B* (2018)

- In-wetting: Droplets at inner leaflet
- Out-wetting: Droplets at outer leaflet
- Droplet view: Wetting morphologies
- Membrane view: Fluid-elastic scaffolding
- Ex 1: Partial wetting, contact angles
- Ex 1: Capillary forces and deformations
- Ex 2: Complete wetting by γ phase
- Ex 2: Endocytosis of β droplet



Example 1

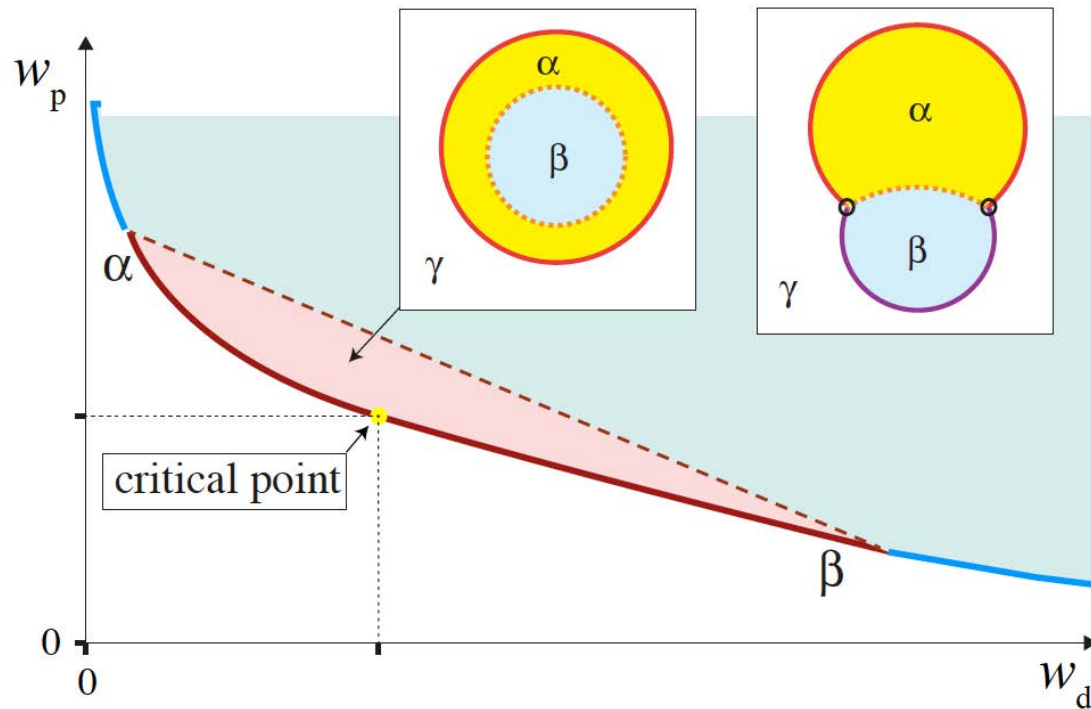


Example 2

Phase Diagram for PEG + Dextran

Liu et al, *Langmuir* (2012)

- Phase diagram of $\alpha = \text{PEG}$ and $\beta = \text{dextran}$:

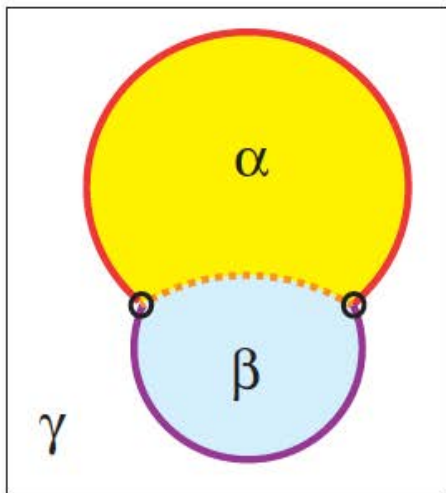


coexistence region
= two subregions:
pink: complete wetting
blue: partial wetting

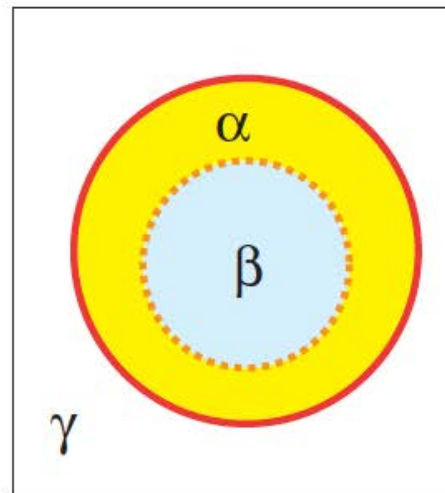
- For all lipid compositions, same ordering of complete and partial wetting subregions
- Partial-to-complete wetting transition at boundary tie line

In-Wetting Morphologies

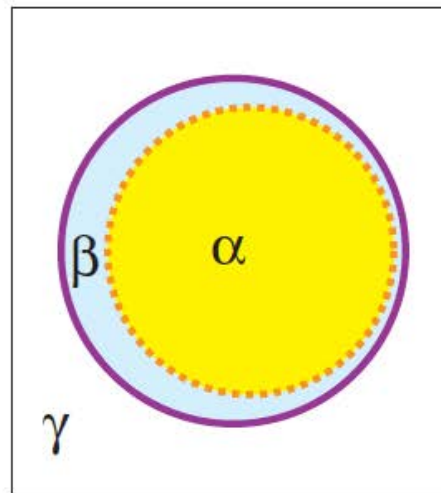
- Three aqueous phases α , β , γ
- Phase coexistence of α and β , γ is ext spectator phase
- GUV membrane encloses α and β



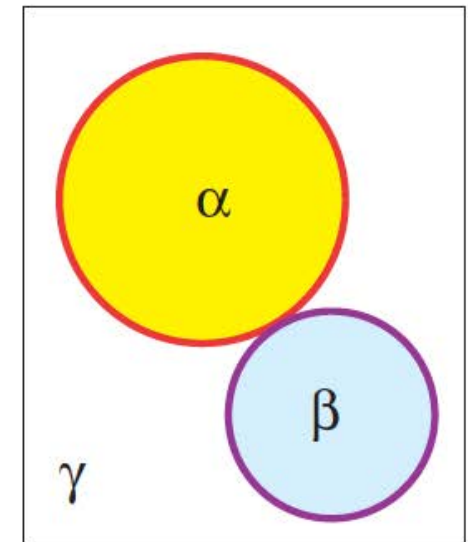
Partial wetting
by α and β



Complete
wetting by α



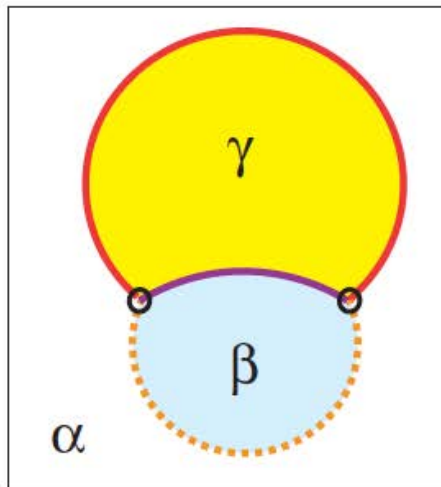
Complete
wetting by β



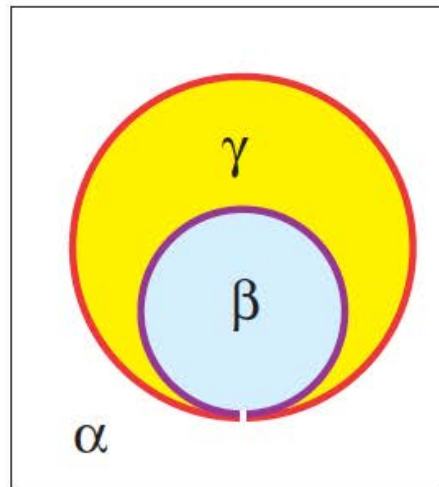
Complete
wetting by γ ,
Membrane neck

Out-Wetting Morphologies

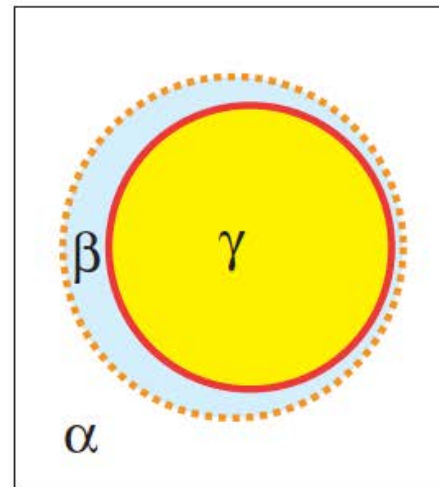
- Three aqueous phases α , β , γ
- Phase coexistence of α and β in exterior solution
- GUV membrane encloses spectator phase γ



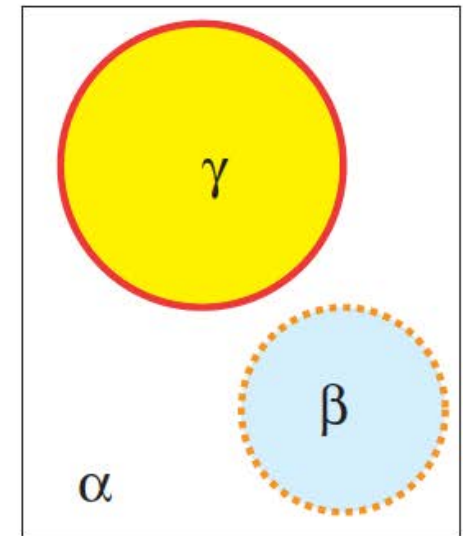
Partial wetting
by α and β



Complete
wetting by γ ,
Membrane neck



Complete
wetting by β



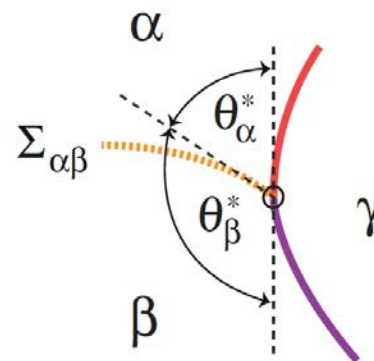
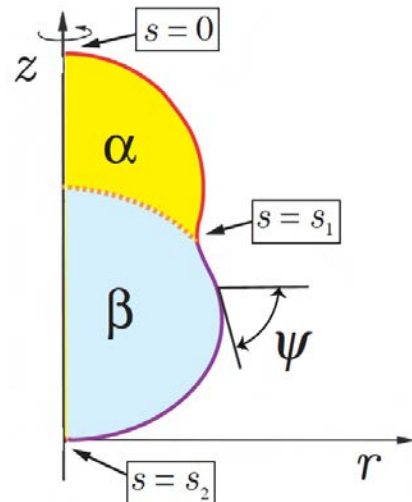
Complete
wetting by α ,
Dewetting

Nanoscopic Scales

RL, *J. Chem. Phys. B* (2018)

- Shapes with smooth bends and no kinks
- Axisymmetric shapes, energy minimization:

arc length s
contact line
at $s = s_1$



common tangent
at contact line,
two intrinsic
contact angles

θ_{α}^* and θ_{β}^* with

$$\theta_{\alpha}^* + \theta_{\beta}^* = \pi$$

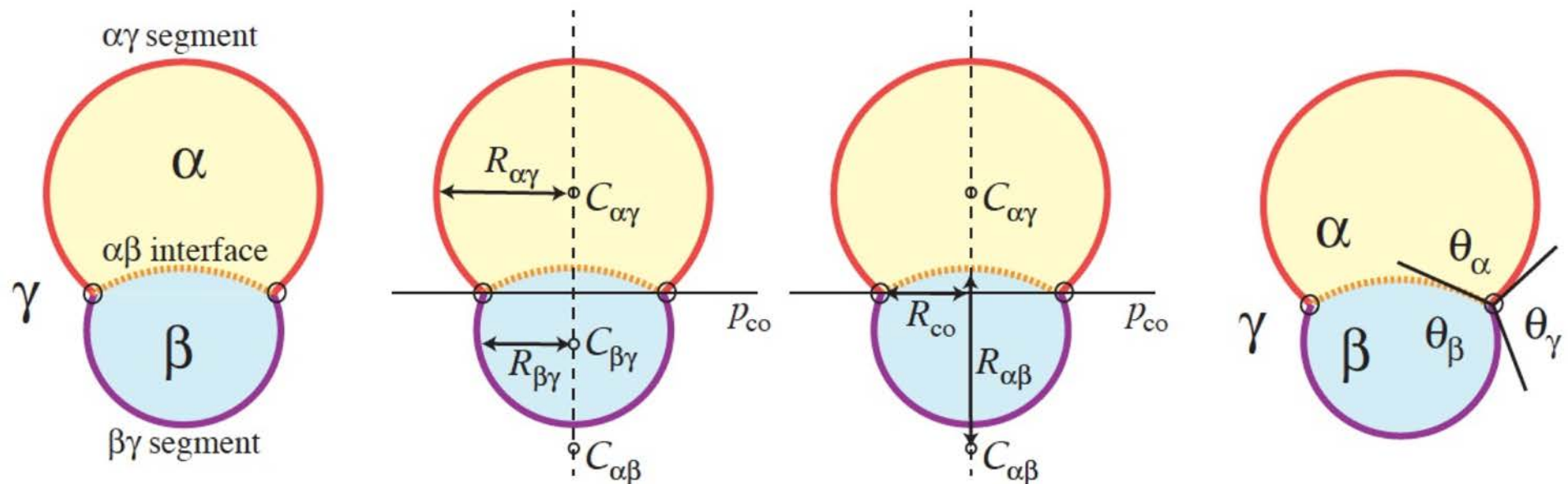
- Complex matching conditions along contact line
- Simplification if $\alpha\gamma$ and $\beta\gamma$ segments have identical curvature elastic properties

Kusumaatmaja et al, *Phys. Rev. Lett.* (2009)

Mesososcopic Scales

RL, *J. Chem. Phys. B* (2018)

- Experimental shapes consist of three spherical caps
- Apparent contact angles θ_α , θ_β , and θ_γ with $\theta_\alpha + \theta_\beta + \theta_\gamma = 2\pi$
- Geometry of three spherical caps, common contact line



- Shape determined by four radii $R_{\alpha\gamma}$, $R_{\beta\gamma}$, $R_{\alpha\beta}$, and R_{co} and by positions of three sphere centers $C_{\alpha\gamma}$, $C_{\beta\gamma}$, and $C_{\alpha\beta}$

Effective Membrane Tensions

- Shape equations for mean curvatures $M = \pm 1/R$
- Balance between pressures and tensions:

$$P_\alpha - P_\beta = 2\Sigma_{\alpha\beta}M_{\alpha\beta} \quad P_j - P_\gamma = 2\Sigma_{j\gamma}^{\text{eff}}M_{j\gamma}$$

- Effective membrane tension:

$$\Sigma_{j\gamma}^{\text{eff}} \equiv \Sigma + W_{j\gamma} + \sigma_{j\gamma} - 2\kappa_{j\gamma}m_{j\gamma}M_{j\gamma}$$

- Overall lateral stress Σ from tensile forces
- Adhesive strength W from attractive molecular forces
- Spontaneous tension σ from bilayer asymmetry
- Curvature dependent term

Tensions and Angles

RL, *J. Chem. Phys. B* (2018)

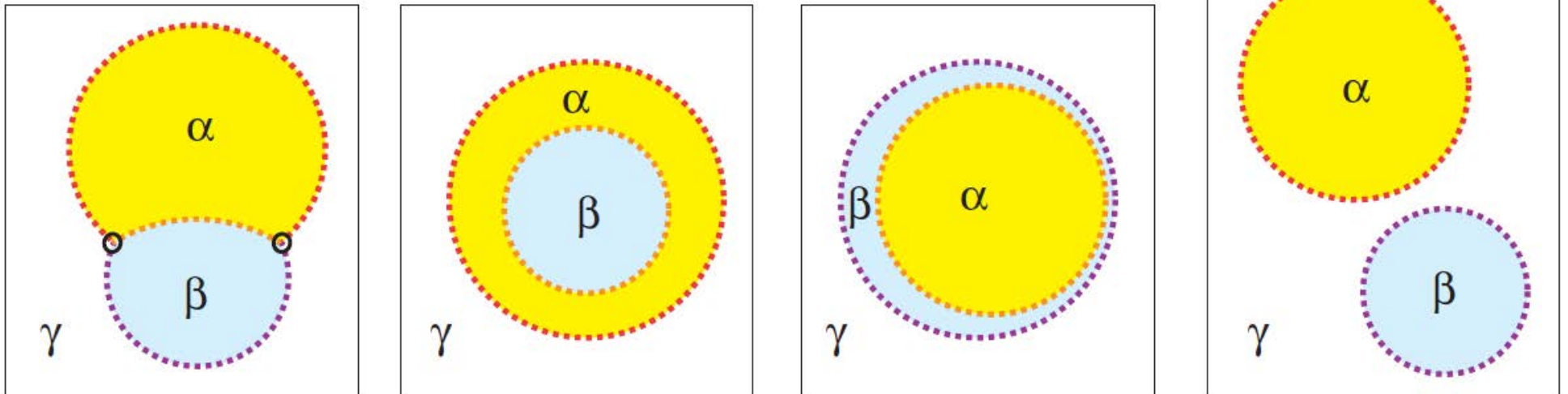
- Combine three shape equations with geometric relation
- Eliminate mean curvature $M_{\alpha\beta}$
- Relation between effective tensions and contact angles:

$$M_{\alpha\gamma} \left(\frac{\Sigma_{\alpha\gamma}^{\text{eff}}}{\Sigma_{\alpha\beta}} - \frac{\sin \theta_{\beta}}{\sin \theta_{\gamma}} \right) = M_{\beta\gamma} \left(\frac{\Sigma_{\beta\gamma}^{\text{eff}}}{\Sigma_{\alpha\beta}} - \frac{\sin \theta_{\alpha}}{\sin \theta_{\gamma}} \right)$$

- Relation depends on mean curvatures $M_{\alpha\gamma}$ and $M_{\beta\gamma}$
- Contact angles can be measured directly
- Complex parameter dependence via effective tensions

Droplets without Membrane

- Three coexisting aqueous phases α , β , γ



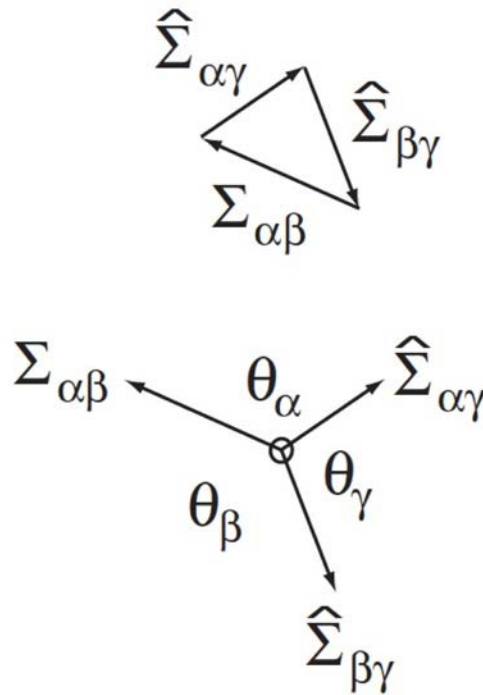
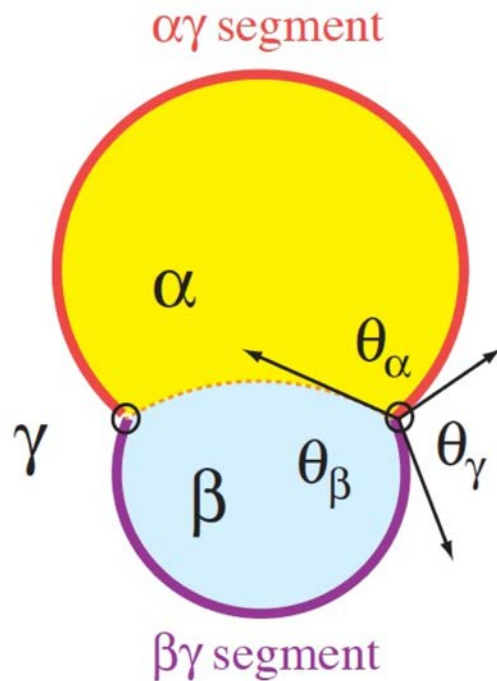
- Similar relation:

$$M_{\alpha\gamma} \left(\frac{\Sigma_{\alpha\gamma}}{\Sigma_{\alpha\beta}} - \frac{\sin \theta_{\beta}}{\sin \theta_{\gamma}} \right) = M_{\beta\gamma} \left(\frac{\Sigma_{\beta\gamma}}{\Sigma_{\alpha\beta}} - \frac{\sin \theta_{\alpha}}{\sin \theta_{\gamma}} \right)$$

- But interfacial tensions Σ_{ij} are material parameters
- Expressions in two parentheses vanish separately

Special Parameter Regimes

- Each segment has small or large spont curvatur
- Balance of tensions at apparent contact line:



Total segment tensions

$$\hat{\Sigma}_{j\gamma} = \Sigma_{j\gamma} + \sigma_{j\gamma}$$

Sum of mechanical and spontaneous tensions

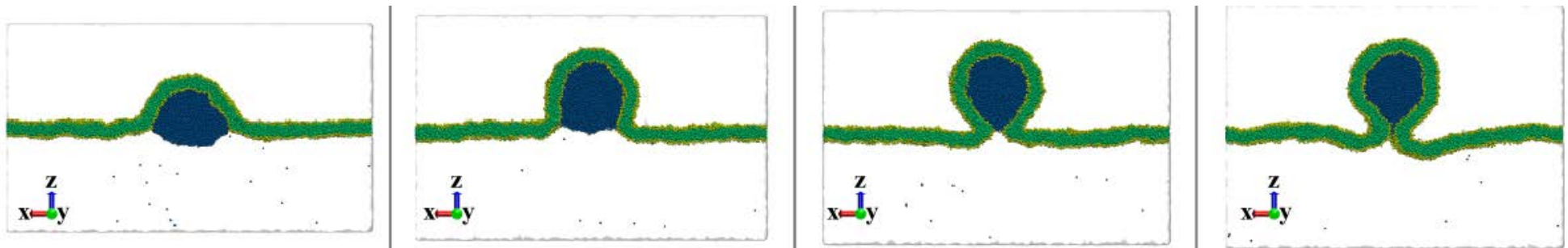
- Balance relations:

$$\frac{\hat{\Sigma}_{\alpha\gamma}}{\Sigma_{\alpha\beta}} = \frac{\sin \theta_{\beta}}{\sin \theta_{\gamma}} \quad \text{and} \quad \frac{\hat{\Sigma}_{\beta\gamma}}{\Sigma_{\alpha\beta}} = \frac{\sin \theta_{\alpha}}{\sin \theta_{\gamma}}$$

Scaffolding Mechanisms

Scaffolding mechanisms and parameters:

- Tensile forces generate overall lateral stress Σ
- Adhesive strength $W < 0$ creates contact area
- Interfacial tension $\Sigma_{\alpha\beta}$ generates capillary forces
- Endocytosis: larger contact area + smaller interfacial area



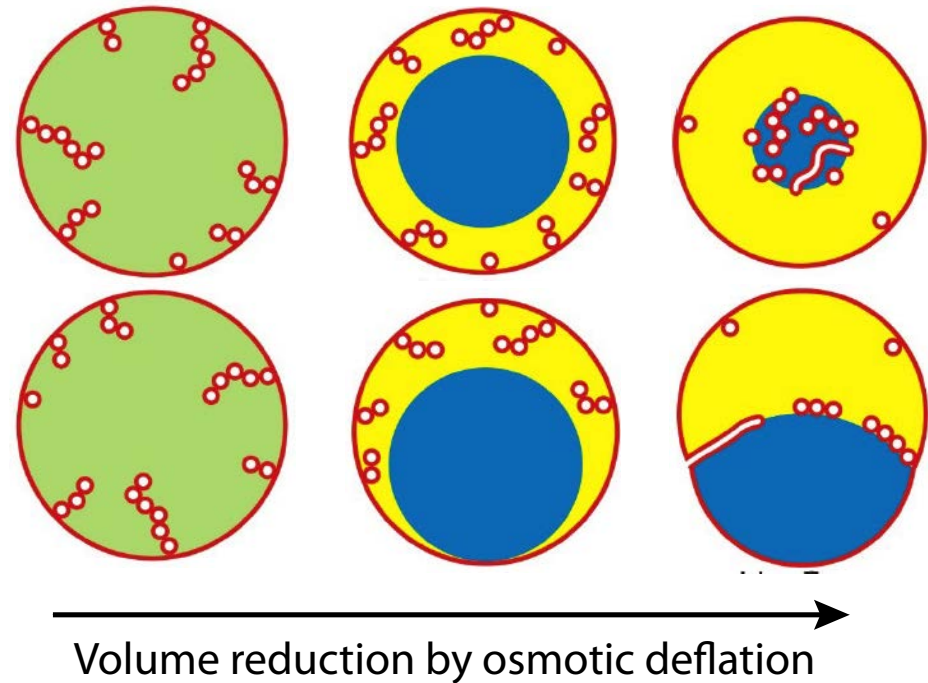
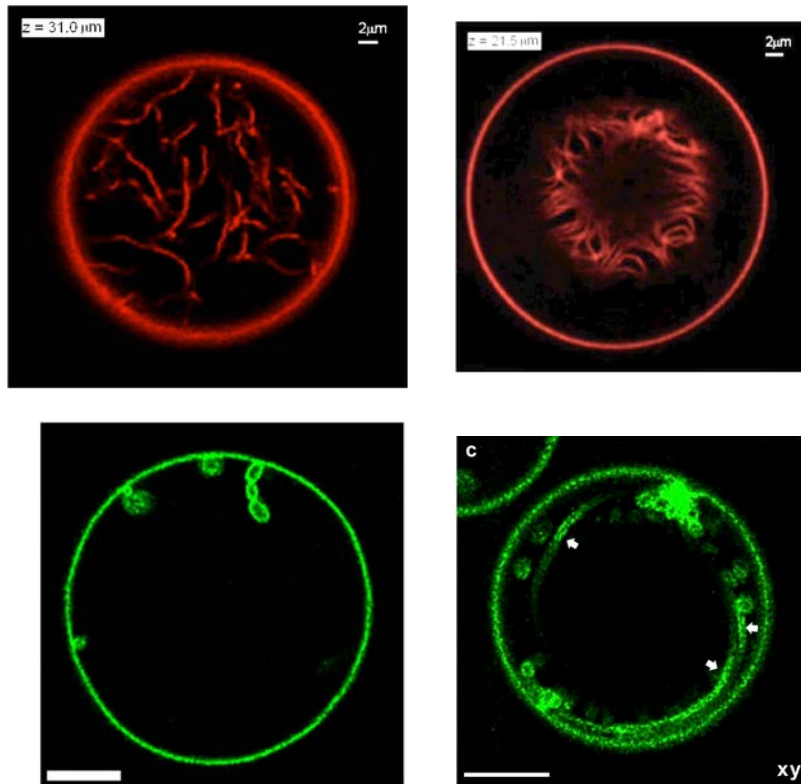
Vahid Satarifard (unpublished)

- Droplet generates bilayer asymmetry and spont curv

Droplet-Induced Curvature

Liu, Agudo ... RL, *ACS Nano* (2015)

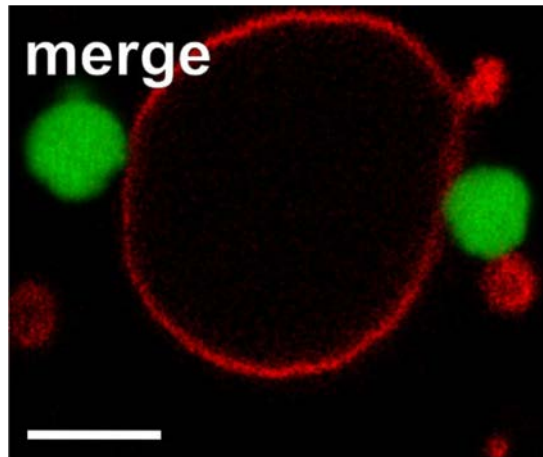
- Tubulation of membrane segment in contact with PEG-rich phase:



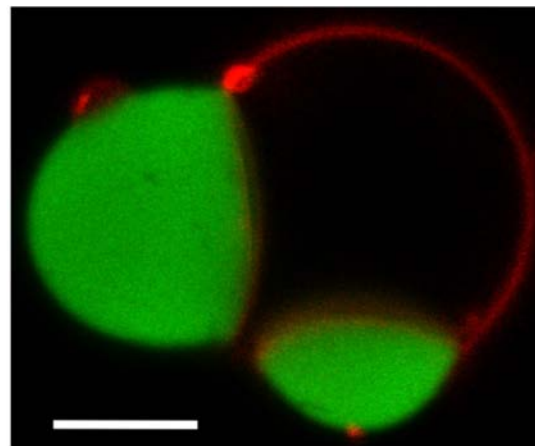
Biomolecular Condensates

Brangwynne ... Hyman, *Science* (2009)

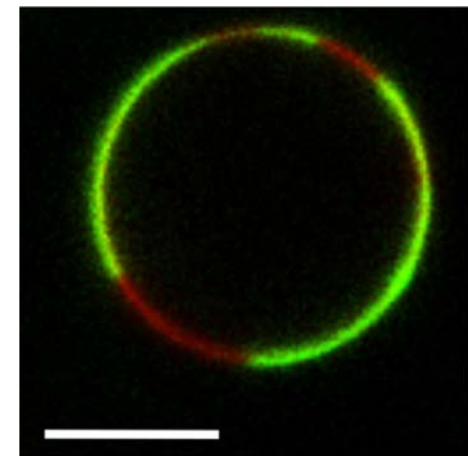
- Biomolecular condensates that behave like liquid droplets
- Enriched in intrinsically disordered proteins (IDPs)
- Example for IDP: RNA-binding protein FUS
- Interaction of FUS-droplets with GUVs, two subsequent wetting transitions:



dewetting for
high salt



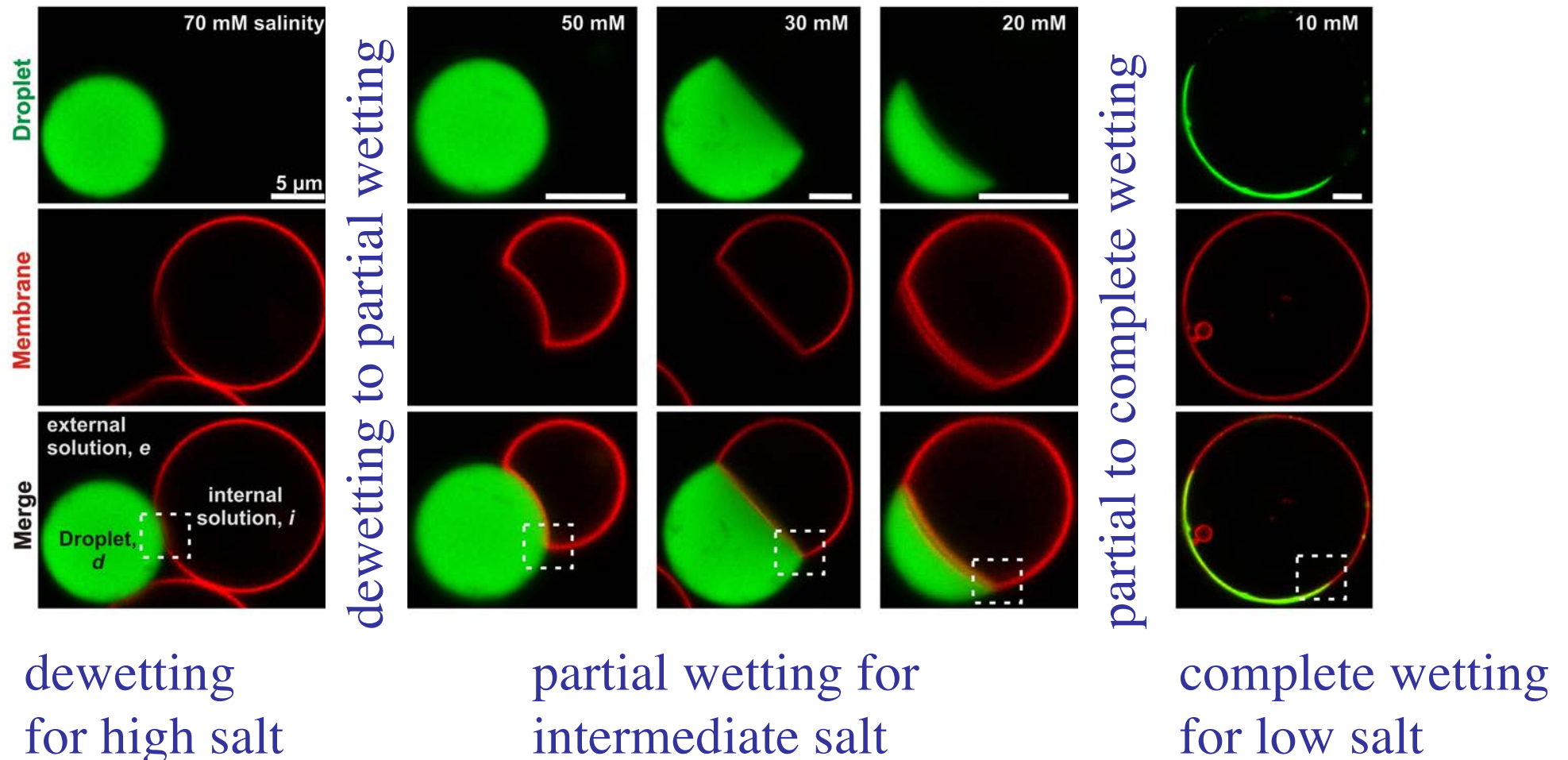
partial wetting for
intermediate salt



complete wetting
for low salt

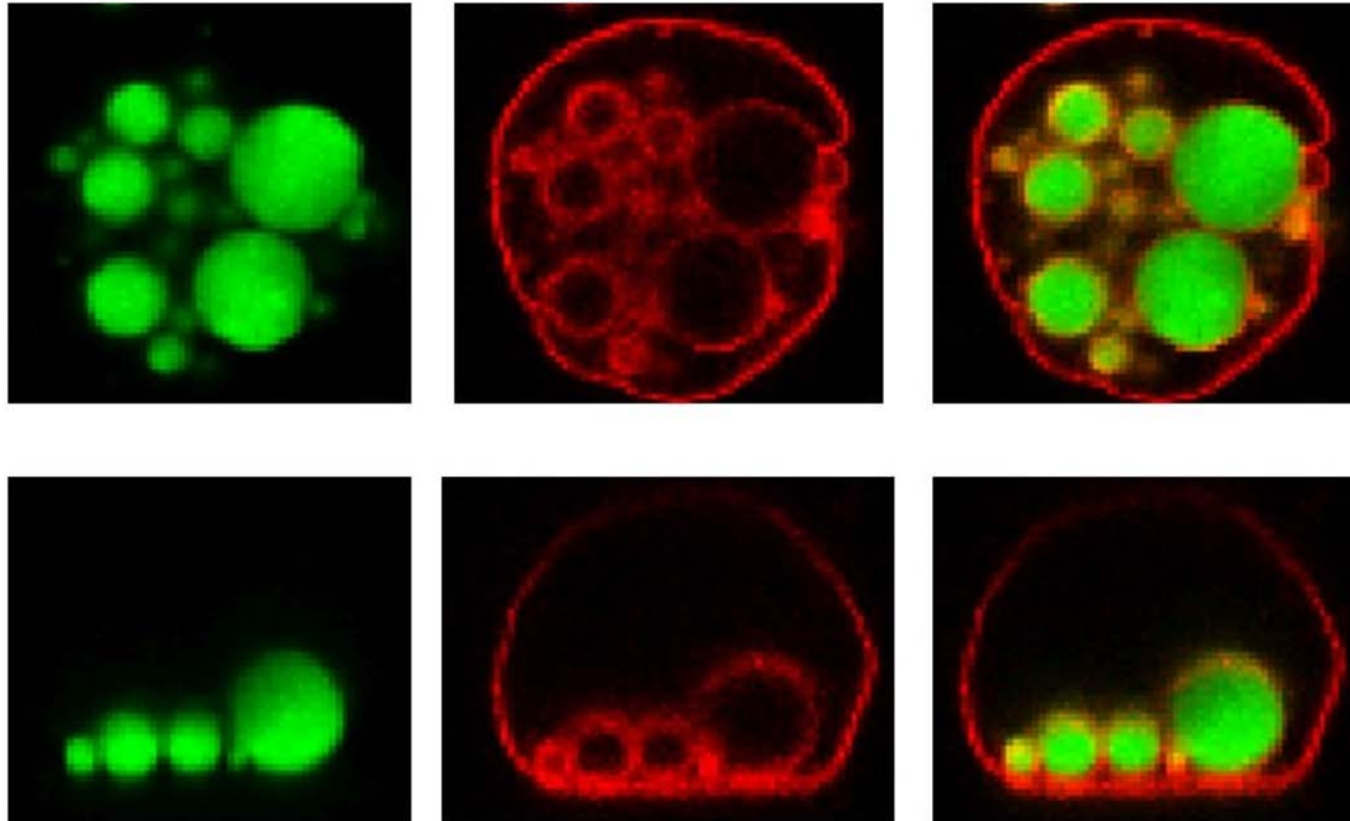
Two Wetting Transitions

- GUV + FUS-rich organelle + salt



Endocytosis of Condensates

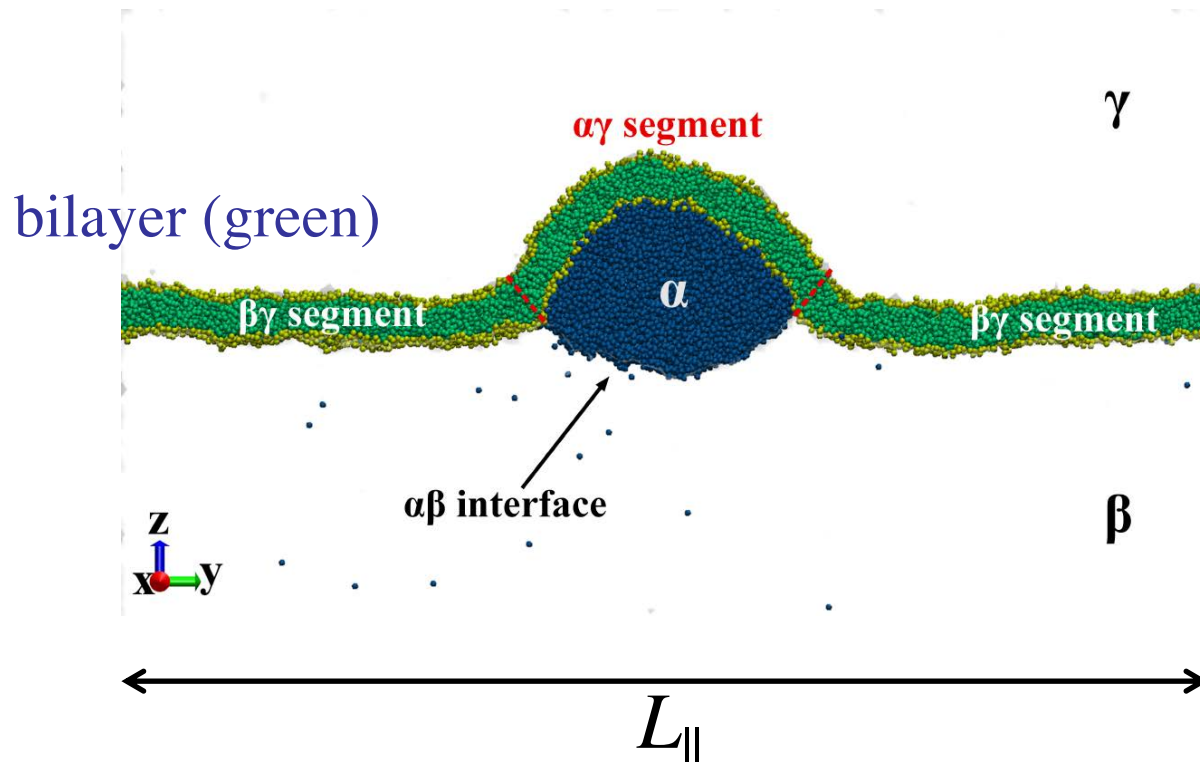
- Green FUS-rich condensates engulfed by red GUV:



Lipid Bilayer + Nanodroplet

Satarifard, Grafmüller, RL (unpublished)

- Molecular simulations of lipid bilayer + nanodroplet
- Lateral box size $L_{||}$ determines mechanical tension
- Mechanical tension \sim size $L_{||}$ as control parameter

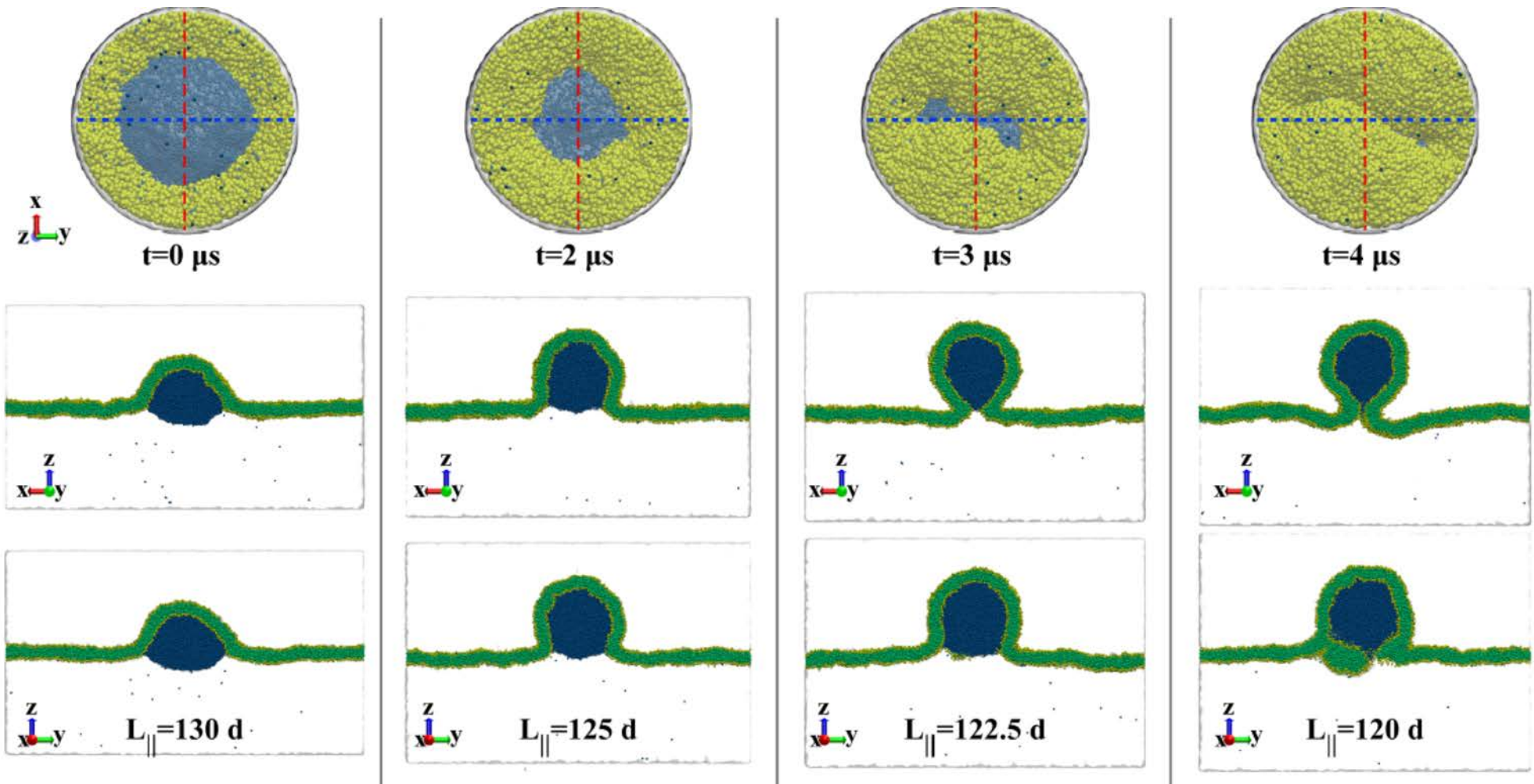


three aqueous
phases α , β , γ

α droplet (blue)
coexists with
 β phase (white)

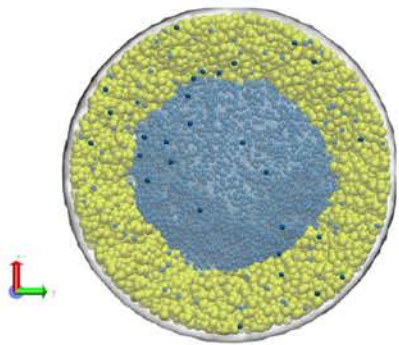
three surface
segments
 $\alpha\beta$, $\alpha\gamma$, $\beta\gamma$

Engulfment from Different Angles

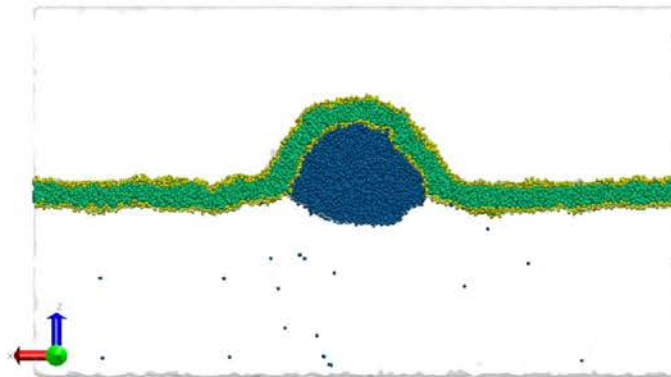


Engulfment Movie

Bottom View



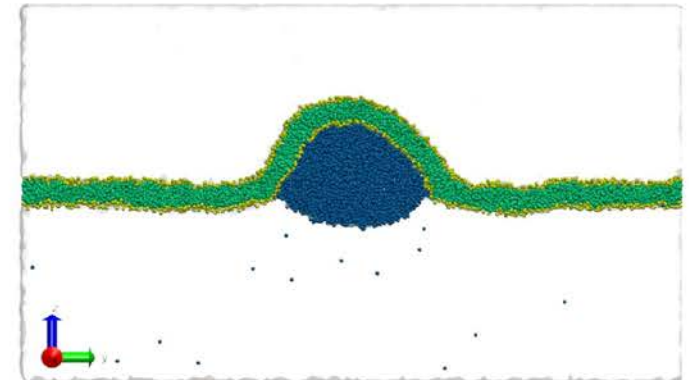
Cross Section 1



$L_{||}=130$ [d]

$t=0$ [μs]

Cross Section 2

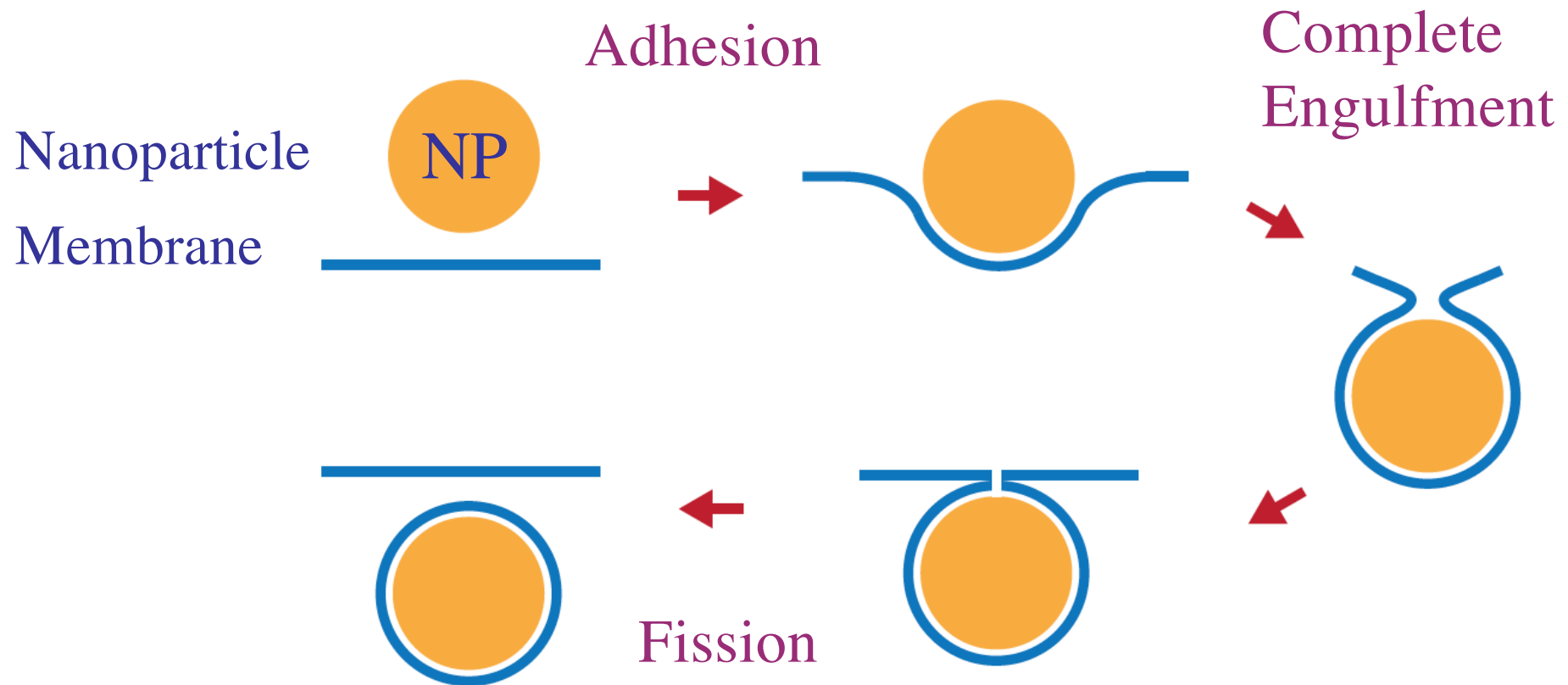


Vahid Satarifard
(unpublished)

- Reduction of size $L_{||}$ from $130 d$ to $120 d$
- Axisymmetry broken for $L_{||}$ around $124 d$
- Caused by negative line tension of contact line

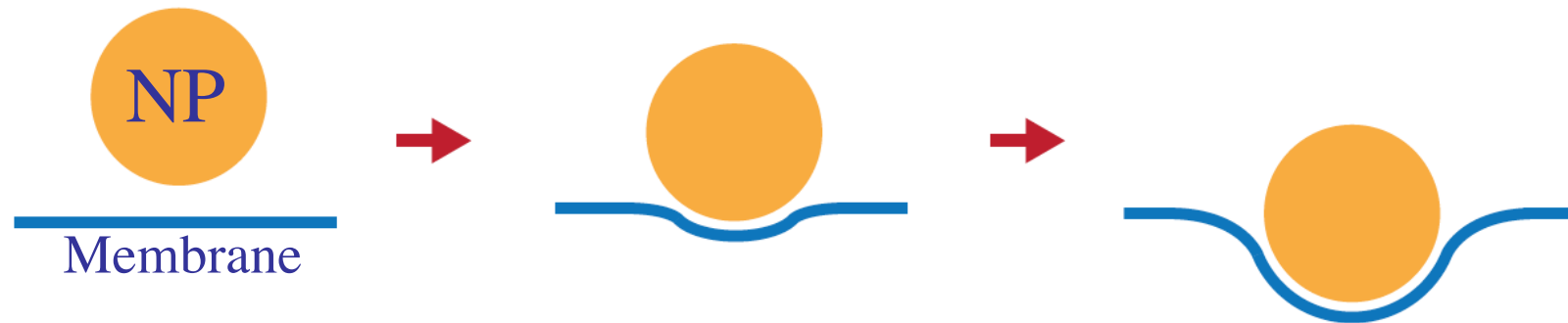
- Intro: Membranes and Giant Vesicles (GUVs)
- GUVs and Aqueous Two-Phase Systems
- Spont Tubulation and Tension
- Morphological Complexity
- Wetting and Fluid-Elastic Scaffolding
- GUVs and Biomolecular Condensates
- Endocytosis of Nanodroplets
- **Endocytosis of Nanoparticles**
- Outlook: Droplet-Stabilized GUVs

Endocytosis of Nanoparticles



- Dissecting endocytosis into three basic steps:
Onset of Adhesion, Complete Engulfment, Fission

Adhesion: Basic Aspects



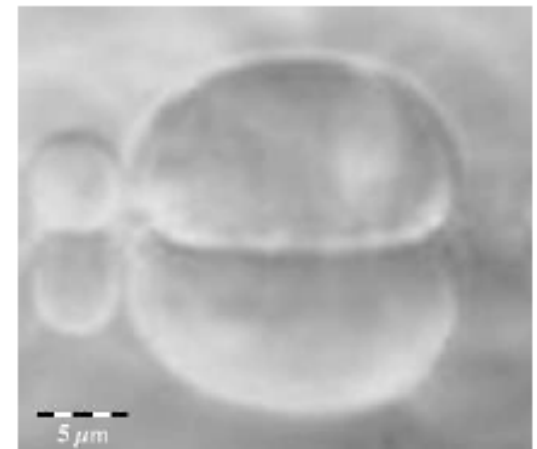
- Attractive interactions between NP and membrane
- Van der Waals, electrostatic, receptor-ligand
- Gain of adhesion free energy but increase of elastic membrane energy
- Competition between adhesion and bending
- Bending rigidity κ versus adhesive strength $|W|$

Adhesion Length

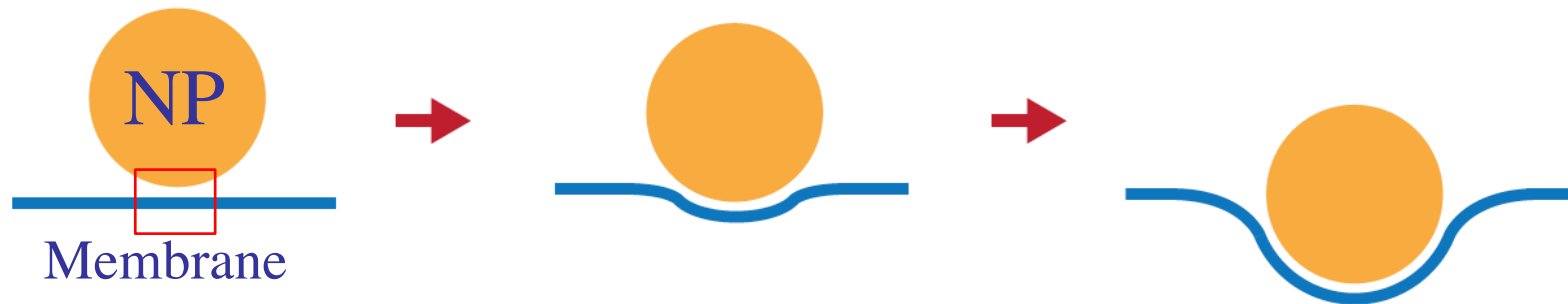
- Adhesive strength $|W|$ = adhesion free energy per area
- Bending rigidity κ and adhesive strength $|W|$ define adhesion length

$$R_W = (2\kappa/|W|)^{1/2}$$

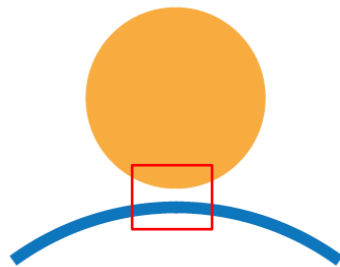
- For specific NP-membrane systems, R_W varies between 10 nm and 3 μm !
- Large R_W values can be measured via membrane curvature along contact line



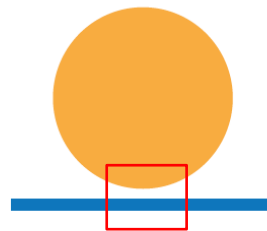
Onset of Adhesion: Key Parameters



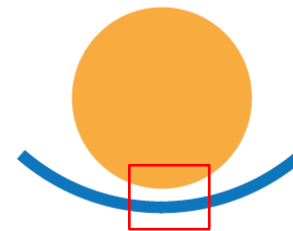
- Three key parameters for onset of adhesion:
Adhesion length R_W , Particle size R_{pa} , and
Membrane curvature M at point of contact
- Membrane curvature M can be positive or negative:



$$M > 0$$



$$M = 0$$



$$M < 0$$

Onset of Adhesion: Local Criterion

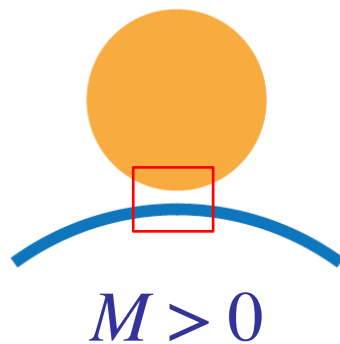
Agudo-Canalejo and RL, *ACS Nano + Nano Letters* (2015)

- Membrane starts to spread over particle if

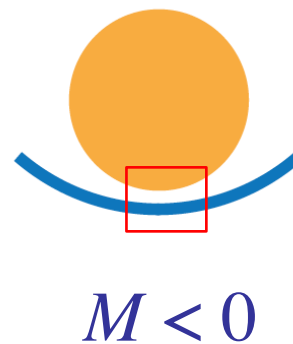
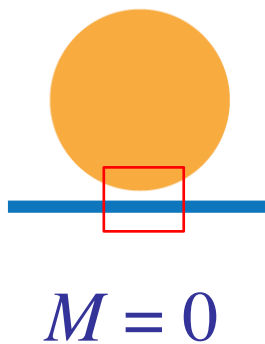
$$M \leq 1/R_W - 1/R_{pa} =: M_{co}$$

contact curvature
 M_{co} is threshold
value for M

- Example: $R_W = R_{pa}$ or $M_{co} = 0$



no adhesion

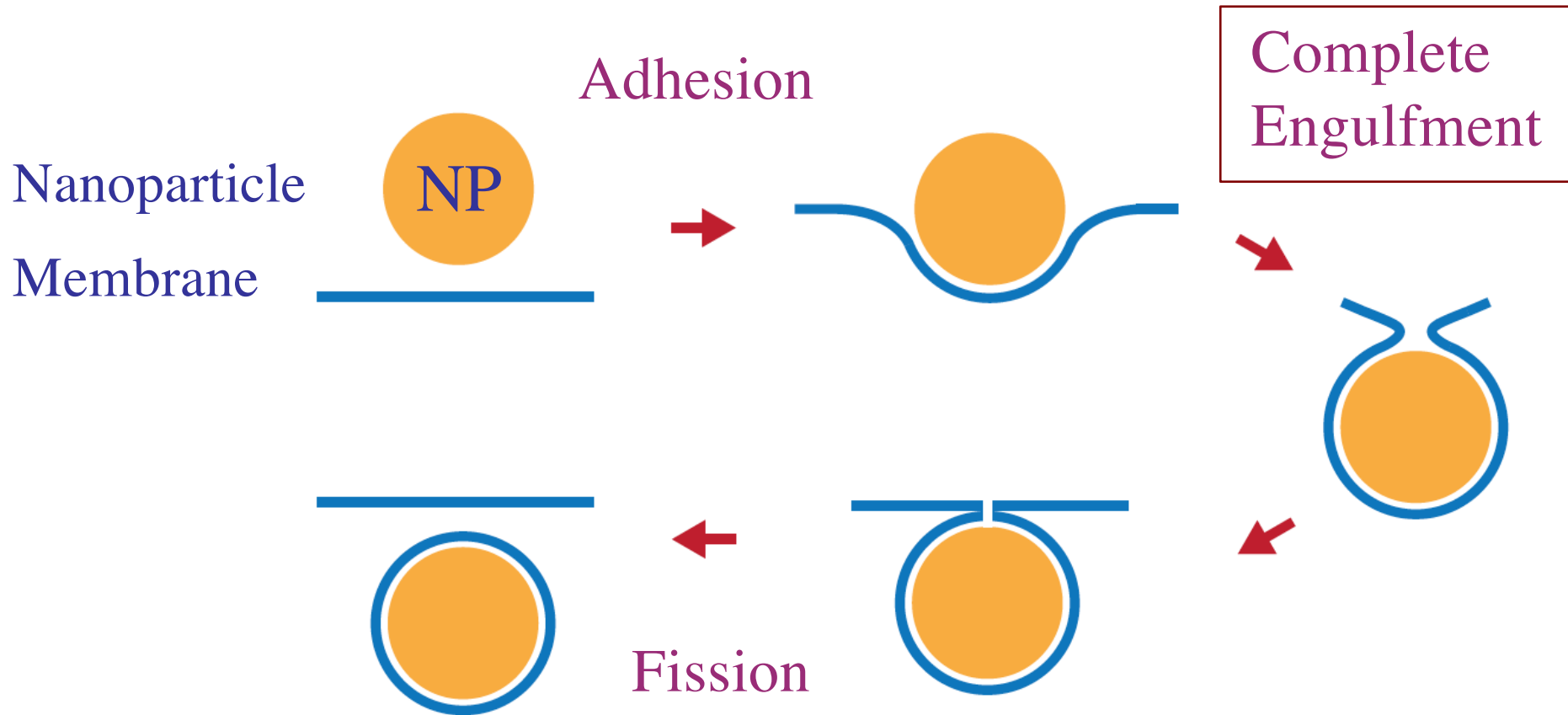


adhesion and spreading

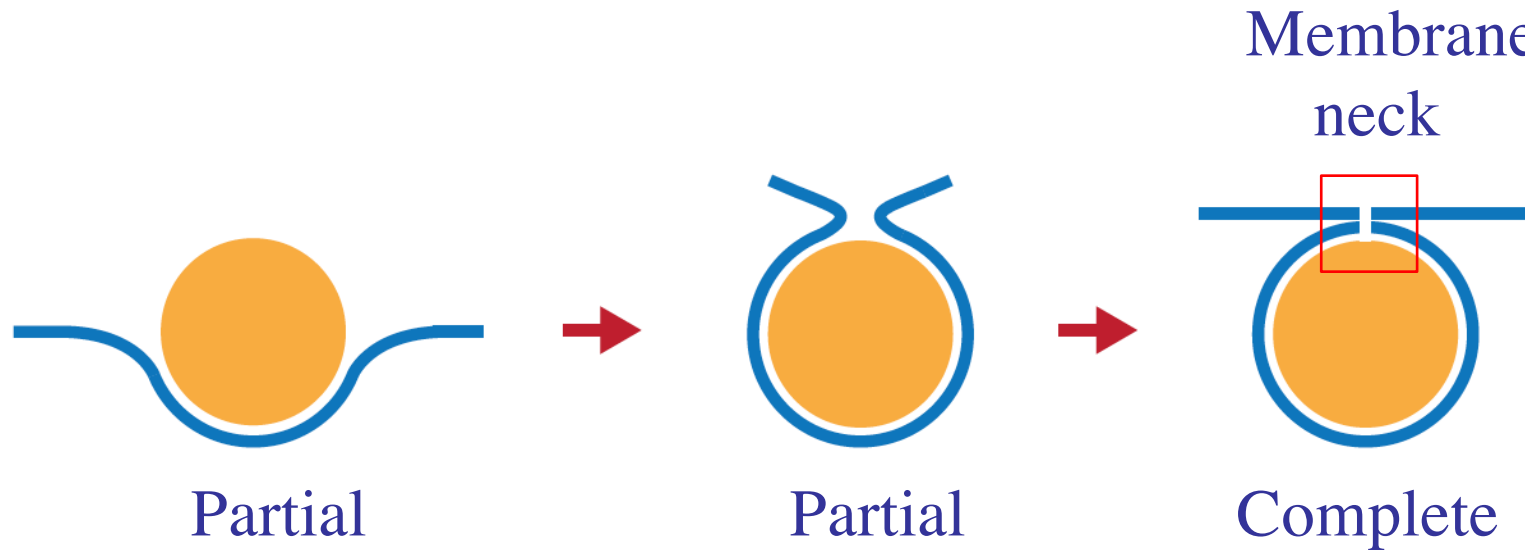


- Large contact curvature M_{co} for small R_W or large $|W|$

Endocytosis: Complete Engulfment



Engulfment: Basic Aspects



- After onset of adhesion, membrane spreads over NP
- Membrane may engulf NP only partially or completely
- Complete engulfment involves closed membrane neck
- Necessary condition for complete engulfment:
Closed membrane neck must be stable

Neck Stability: Local Criterion

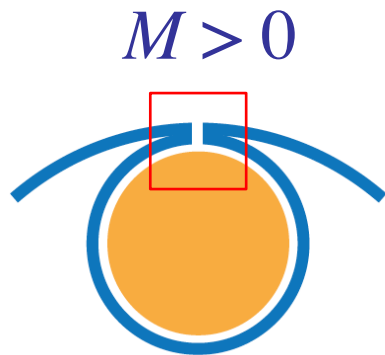
Agudo-Canalejo and RL, *ACS Nano + Nano Letters* (2015)

- Closed membrane neck is stable if membrane curvature

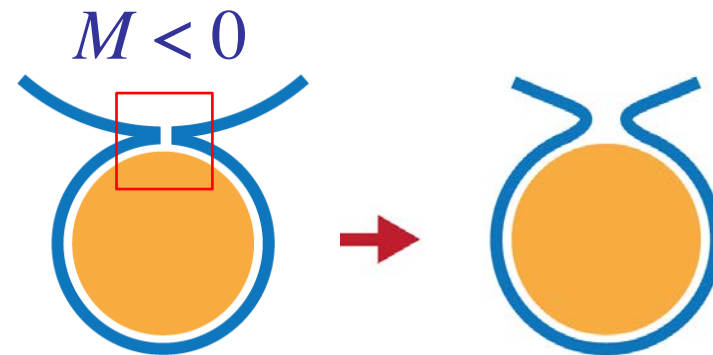
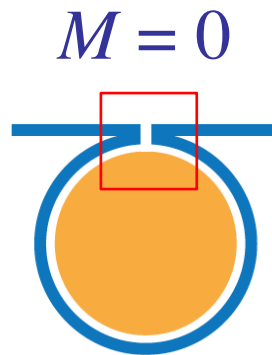
$$M \geq 2m + 1/R_{pa} - 1/R_W =: M_{ne}$$

2nd threshold
value for M

- Example: $M_{ne} = 2m + 1/R_{pa} - 1/R_W = 0$

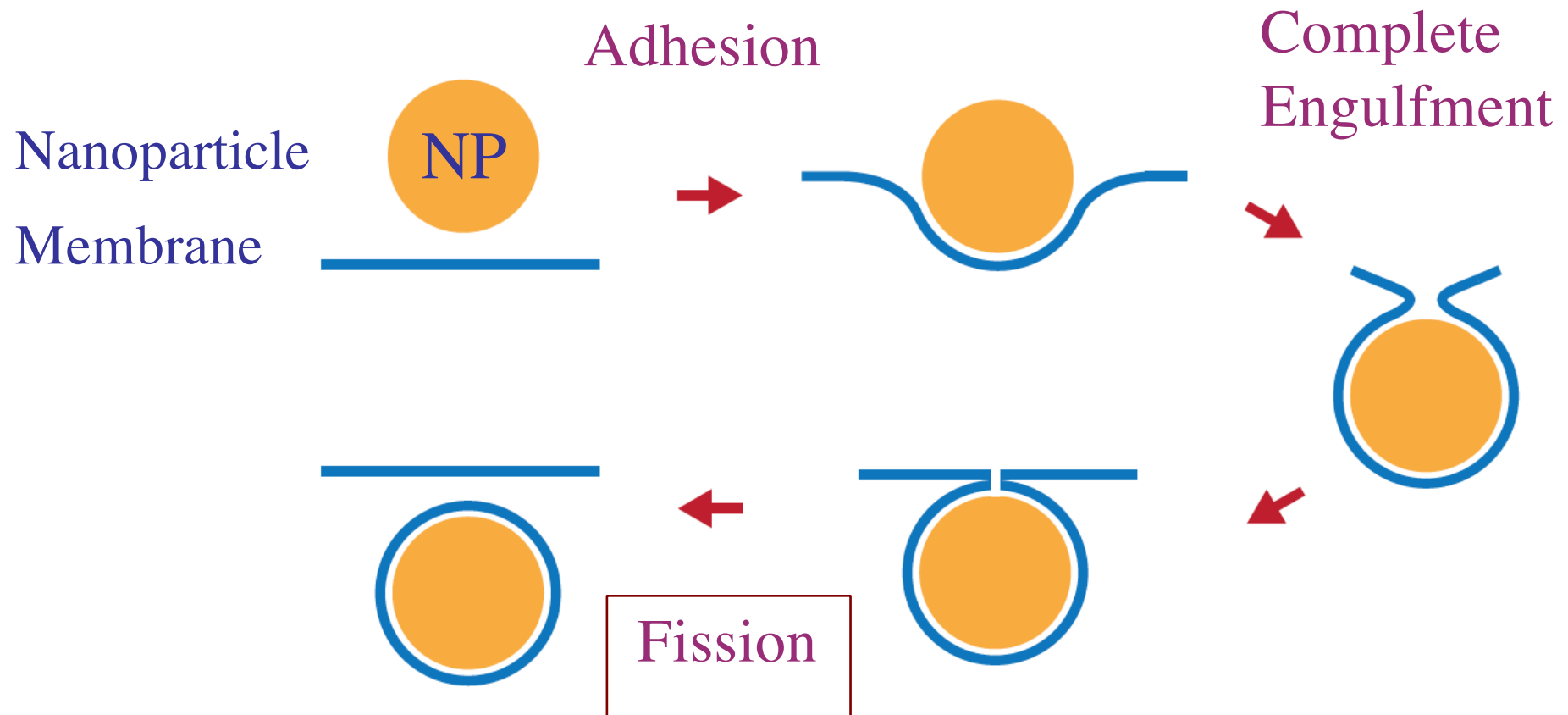


closed neck is stable



closed neck is unstable
and opens up

Endocytosis: Fission



Effective Constriction Force

- Closed neck stability with force f :

$$M + M_{co} - 2m + f (4\pi\kappa)^{-1} \geq 0$$

- Effective constriction force

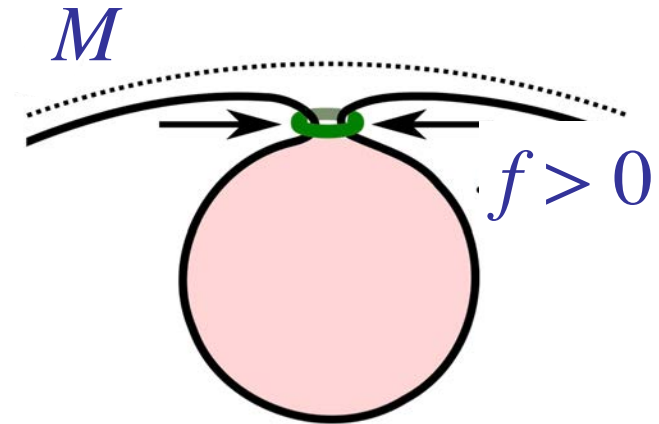
$$f_{\text{eff}} = f + 4\pi\kappa (M + M_{co} - 2m) \geq 0$$

- Engulfment force $f_{\text{eng}} = 4\pi\kappa (M + M_{co} - 2m)$

- Example: $\kappa = 4 \times 10^{-19}$ J, $M = M_{co} = 0$, $m = -1/(100 \text{ nm})$

$$\Rightarrow \text{Engulfment force } f_{\text{eng}} = 100 \text{ pN}$$

- Sufficient to create two hydrophobic bilayer edges



Receptor-Mediated Endocytosis

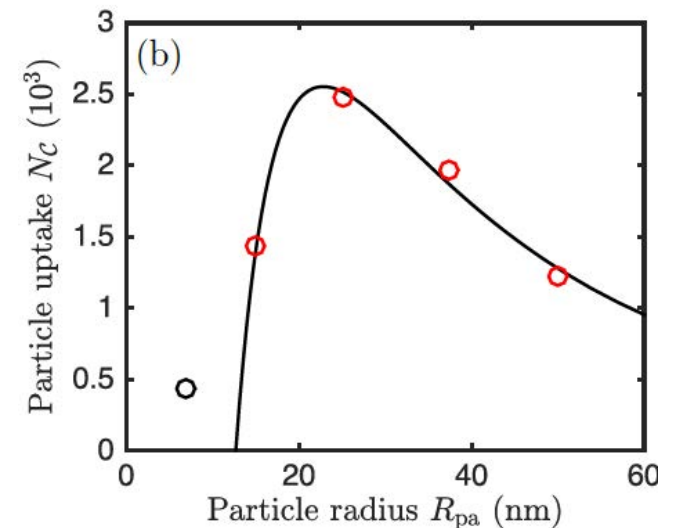
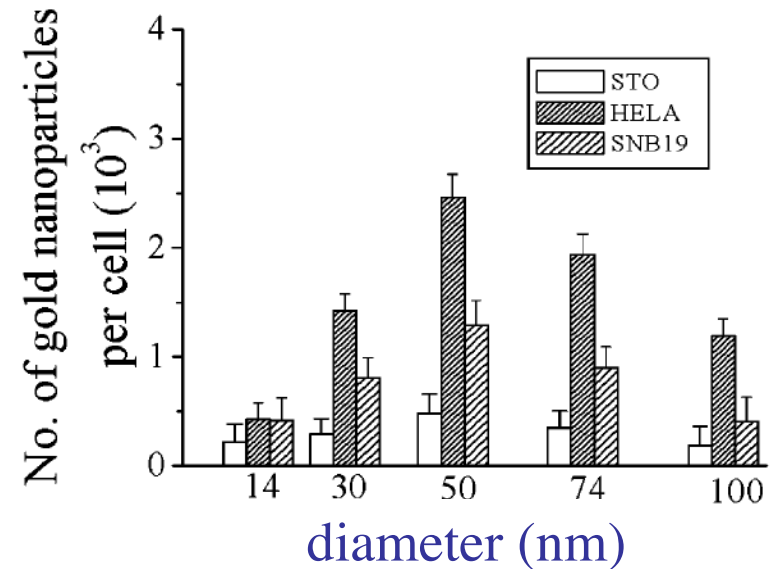
Chithrani et al, *Nano Letters* (2007)

- Uptake of gold nanoparticles by cells
- Particles bind to transferrin receptors
- Assembly of clathrin-coated vesicles

Non-monotonic size-dependence !

- Cell membrane with two types of segments, bound and unbound
- Bound segment contains protein coat with spont curv $m_{bo} = -1/(40 \text{ nm})$
- Good agreement with exp data:

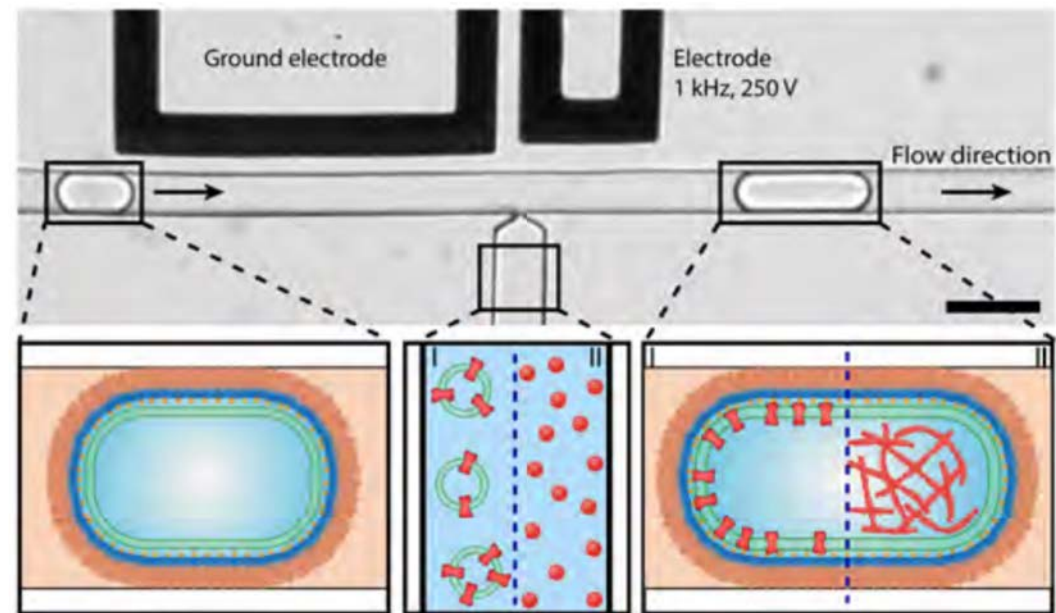
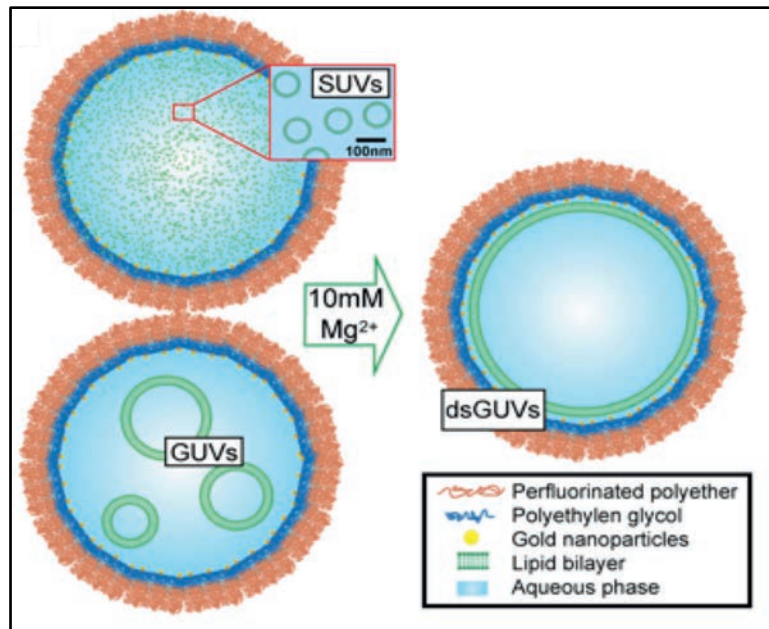
Agudo-Canalejo, RL: *ACS Nano* (2015)



Outlook: Droplet-Stabilized GUVs

Weiss et al, *Nature Materials* (2018)

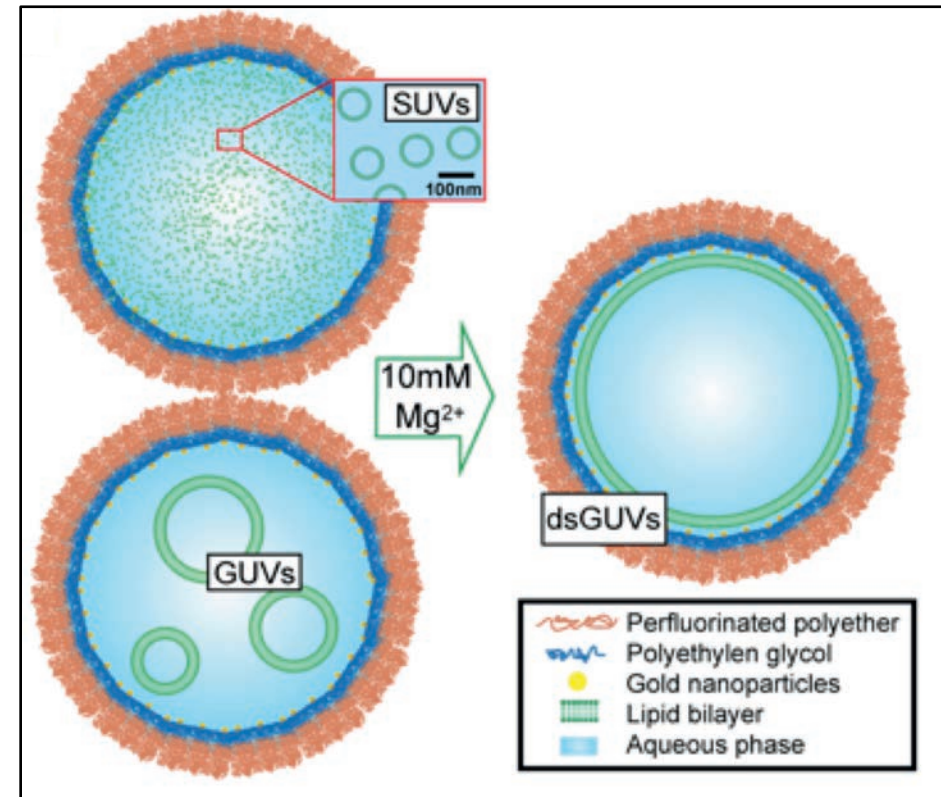
- Four MPIs within MaxSynBio, leading PI: J. Spatz



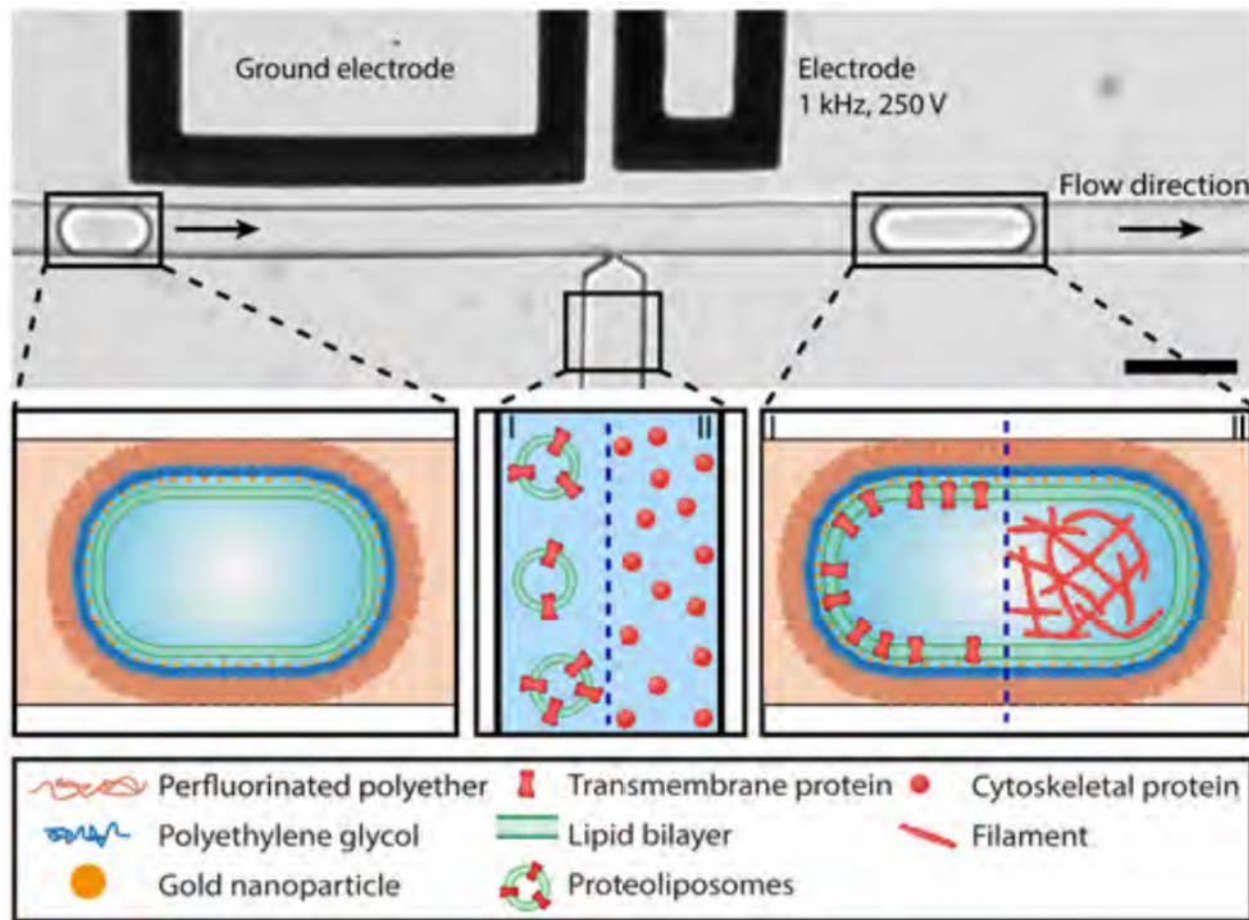
- Water-in-Oil emulsion droplets
- Formation of GUV supported by the droplet surface
- Additional components by pico-injection
- Example: ATP synthase

GUVs within W/O Emulsion Droplets

- Emulsion w/o droplet stabilized by surfactant
- Pico-Injection of small vesicles
- Pico-Injection of Mg^{++}
- Adhesion of vesicles to surfactant layer
- Rupture of vesicles
- Fusion of fragments
=> Formation of a GUV supported by surfactant layer
- Release of encaged GUV from droplet



Sequential Pico-Injections



- Pico-injection of membrane and cytoskeletal proteins
- Incorporation of functional ATP Synthase (FoF1-complex)

Coworkers



Rumiana
Dimova



Vahid
Satarifard



Andrea
Grafmüller

- Collaborations:

Joachim Spatz, Tony Hyman, Titus Franzmann



Jaime Agudo-C.