Morphological Complexity of Biomembranes and Vesicles

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- Membranes and vesicles
- Membrane tension and curvature elasticity
- Spontaneous curvature and membrane necks
- Morphological complexity

"Physical understanding from fruitful interplay between theory, simulation, and experiment"



- Vesicles and Membranes
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Multiscale Membranes



• Ani

• Animal cell

Morphology of Vesicles

- Giant unilamellar vesicles (GUVs), tens of micrometers
- Remodelling in response to various perturbations:



Nanotubes from polymer adsorption, tube width ~ 100 nm Formation of intramembrane domains, 2D phase separation Small buds from protein adsorption, bud size $\sim \mu m$ Remodelling by adhering or partially wetting droplets

• What determines these strongly nonspherical shapes of membranes and vesicles?

Membrane Fluidity

• Fluid membranes, i.e., fast lateral diffusion:

Diffusion constant ~ $\mu m^2/s$

• Lateral diffusion => demixing and domain formation



lipid swapping ~ ns



 Fluidity => Flexibility
 Formation of spherical bud (with Gaussian curvature) => Direct evidence for fluidity



40 μm 5

Membranes as Flexible Sheets

- Membrane as thin elastic sheet
- Elastic Deformations

Stretching \rightarrow Shearing \rightarrow Bending \rightarrow

• Fluid Membranes

Hardly stretchable (without rupture)

Shear -> Flow

Main contribution related to bending and curvature

Emergence of Curvature

• Fluid membranes as smooth surfaces: Fluidity => no intrinsic coordinates Invariance under coordinate transformations Principal curvatures C_1 and C_2 Mean curvature $M = (C_1 + C_2)/2$ Gaussian curvature $G = C_1 C_2$ Canham



Canham, Helfrich, E. Evans (1970s)

- Mesoscopic description, ignores molecular architecture
- Emergence of curvature on nanoscopic scales:

Bending undulations for membrane patches > 6 nm



• Vesicles and Membranes



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For Comparison: Liquid Droplets

- \bullet Coexistence of two liquid phases α and β
- Shape energy of a β droplet within the bulk α phase:

$$E = V_{\beta} (P_{\alpha} - P_{\beta}) + A_{\alpha\beta} \Sigma_{\alpha\beta}$$



Two parameters: Pressure difference P_{β} - P_{α} and interfacial tension $\Sigma_{\alpha\beta}$

• First variation of shape energy leads to Laplace equation for mean curvature *M* :

$$2 M \Sigma_{\alpha\beta} = P_{\beta} - P_{\alpha} = \Delta P$$

Interfacial tension is constant (for constant temperature)
 => Interface attains shape with constant M

Shape Energy of Vesicles

• Shape energy of a vesicle

Deuling, Helfrich, J. Physique (1976)

 $E = -\Delta P V + \Sigma A + \text{curvature-elastic terms}$

- First two terms have the same form as for droplets
- $\Delta P = P_{in} P_{ex}$ is again the pressure difference
- But the area term has a different meaning: in contrast to droplets, the membrane area A is essentially fixed and the tension Σ can either be viewed as a Lagrange multiplier to prescribe the area or as the elastic stress acting to stretch the weakly compressible membrane RL, Adv. Colloid Interface Sci. (2014)
- => While the interfacial tension $\Sigma_{\alpha\beta}$ represents a material parameter, the membrane tension Σ is shape-dependent

Curvature Elasticity

Canham (1970) Helfrich (1973) Seifert, Berndl, RL (1991)

- Local mean curvature *M* tries to adapt to spontaneous (or preferred) curvature *m*
- Curvature or bending energy:

$$E_{cu} = \int dA \ 2 \ \kappa (M - m)^2$$



11

integral over membrane area A

- 2nd fluid-elastic parameter: Bending rigidity κ Dimensions of energy, $\kappa = 10^{-19} \text{ J} = 20 \text{ k}_{\text{B}} \text{ T}$
- Range of spontaneous curvatures *m* from 1/(20 nm) to 1/(20 μm)

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Spontaneous or Preferred Curvature

- Spontaneous or preferred curvature *m* describes bilayer asymmetry = asymmetry between two leaflets
- Different molecular mechanisms for bilayer asymmetry:



Asymmetric composition, e.g., ganglioside

Asymmetric adsorption of small molecules Asymmetric protein coats, e.g. BAR-domain

Local Curvature Generation

• Local curvature generated on nanoscopic scales:



Sign of (Spontaneous) Curvature

- Mean curvature *M* and spontaneous curvature *m* can be positive or negative
- Sign defined with respect to interior/exterior compartments = with respect to inner/outer leaflet

exterior compartment outer leaflet



interior compartment

Mean curvature M is positive (negative) if membrane

bulges towards exterior (interior) compartment 15

inner leaflet

(1+1)-Spheres and Membrane Necks

• Positive sp-curvature *m* > 0 leads to spherical membrane segments connected by closed membrane necks :





- Sphere radii R_1 and R_2
- Mean curvatures $M_1 = 1/R_1$ and $M_2 = 1/R_2$
- Neck curvature $M_{\rm ne} = (1/2) (M_1 + M_2)$
- Closed neck is stable if $0 < M_{ne} \le m$
- Local relation between geometry and material parameter

Stability Regime of (1+1)-Spheres

- Vesicle area A, length scale = vesicle size $R_{ve} = [A/(4\pi)]^{1/2}$
- Dimensionless volume $v \sim V/R_{ve}^{3}$ with $0 < v \le 1$
- Dimensionless sp-curvature $mR_{ve} > 0$



Membrane Necks of Nanovesicles

Rikhia Ghosh, Vahid Satarifard, A. Grafmüller, RL. Nano Letters (2019)

- Molecular dynamics simulations
- Spherical nanovesicle with diameter of 40 nm
- Decreasing vesicle volume v, corresponding to deflation
- Formation of dumbbell with closed membrane neck:



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- Large sp-curvature: Membrane nanotubes Controlled sp-curvature and GUV division

Large Sp-Curvature: Nanotubes

Li ... Dimova, PNAS (2011) Liu ... RL, ACS Nano (2016)

- Lipid mixture of DOPC, DPPC, cholesterol
- Small amounts of fluorescently labeled lipids
- Liquid-disordered (red) and liquid-ordered phase (green)



- Spontaneous tube formation without external forces
- Complex patterns of nanotubes

Sp-Tubulation and Sp-Tension

RL, Faraday Discuss. (2013)

- Tubulation leads to tense mother vesicle
- Total tension in Euler-Lagrange equation has two components:

$$\Sigma = \Sigma + \sigma$$



Mechanical tension Σ stretches the membrane Spontaneous tension $\sigma = 2 \kappa m^2$ for $M \ll m$

- Presence of nanotubes implies dominance of spontaneous tension, mechanical tension can be ignored
- Example: Spont curvature $\approx -1/(100 \text{ nm})$

Spontaneous tension $\sigma \approx 10^{-2}$ mN/m Mechanical tension $\Sigma \approx 10^{-4}$ mN/m

Robustness of tubulated GUVs

- Conventional GUVs: Membrane rupture under osmotic inflation, strong adhesion, micropipette aspiration, ...
- Membrane nanotubes provide area reservoir
- Tubulated GUVs have very low mechanical tension and do not rupture under strong mechanical perturbations
- Robustness demonstrated for inflation and aspiration
- Membrane tension dominated by sp-tension $\sigma = 2 \kappa m^2$
- Mother vesicle behaves like liquid droplet with interfacial tension = sp-tension of membrane

Bhatia ... RL: ACS Nano (2018)



How Do Nanotubes Form?

Liu et al, ACS Nano (2016)

- Tubulation intimately related to budding
- Osmotic deflation of spherical GUV
- Small deflation step leads to single bud



- Bud and mother vesicle connected by membrane neck
- Bud acts as nucleation site for necklace-like tube
- Several pathways for subsequent deflation steps:
 - Formation of new bud
 - Bud into 2-necklace
 - N-necklace into (N+1)-necklace

Nucleation and Growth of Tubes

Liu et al, ACS Nano (2016)

RL, J. Phys. D (2018)

- Spherical GUV, large spont curv *m*
- Osmotic deflation of GUV in discrete steps
- At each step, different morphological pathways:



Morph Complexity: Theory

• After 6th step, 11 morphologies with 6 spherules:



- All beads are connected by membrane necks
- All morphologies have the same area, volume, and curvature energy
- Rugged energy landscape contains 11 intersecting branches
- For large N, number of N-spherule morph grows as $\exp[c \sqrt{N}]_{25}$

Morph Complexity: Experiment

Bhatia, Christ, Steinkühler, Dimova, RL, Soft Matter (under revision)

- GUVs exposed to two simple sugars, sucrose and glucose
- Sugar asymmetry: sucrose inside, glucose outside
- Singe GUV membrane forms N_l large and N_s small spheres connected by closed membrane necks
- Multispheres with $N = N_l + N_s$ and $2 \le N \le 4$:



Multispheres with $N = 1 + N_s$

• One large and N_s small spheres, linear and branched chains:



• Individual small spheres are surprisingly mobile !

N_{*} Equally Sized Spheres

100



(e)

 $N_* = 15$





28

- Vesicles and Membranes
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- Morphological complexity of vesicles Large sp-curvature: Membrane nanotubes



Controlled sp-curvature and GUV division

Controlled Variation of Sp-Curvature

Steinkühler, ..., Wegner, Dimova, RL, Nature Comm (under revision)

• Binding of GFP to certain anchor lipids:



His-tagged GFP NTA-lipids

- GFP solution concentration X as control parameter
- Density Γ of bound GFP increases linearly with X
- Sp-curvature m increases linearly with Γ
- Dilute regime: separation of bound GFPs much larger than lateral size of GFP

Controlled Division of GUVs

• Osmotic deflation + GFP binding

01:09

• Osmotic deflation: Spherical GUV -> dumbbell GUV Increase in GFP -> Neck cleavage -> Two daughter GUVs



Adsorption of GFP onto GUV membrane Deflation leads to dumbbell with membrane neck

Directly after neck cleavage

+GFP-HIS

07:27

+GFP-HIS <u>5 μm</u> Complete division into two smaller GUVs

07:41

Constriction Force from Sp-Curvature

RL, in The Giant Vesicle Book. ed by Dimova and Marques (2019)

• Sp-curvature generates constriction force

 $f = 8\pi \kappa (m - M_{\rm ne})$

acting radially on closed membrane neck:

• Force increases with increasing sp-curvature:





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- Outlook on related topics:



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Bilayer membranes and leaflet tensions Endocytosis of nanoparticles by membranes Wetting and engulfment of droplets

Membrane Tension

- Uniform membranes: Laterally uniform composition and spatially uniform environment
- Uniform membranes experience mechanical tension Σ and spontaneous tension $\sigma = 2 \kappa m^2$
- These two tensions act on the whole membrane
- Optimal packing of lipids ⇔ low mech tension Σ
 ⇔ tensionless membrane
- But biomembranes are molecular bilayers
- What about tensions within individual leaflets?

Bilayer Membranes and Leaflet Tensions

- Bilayer with two leaflets:
 - Two leaflet tensions Σ_1 and Σ_2 with $\Sigma_1 + \Sigma_2 = \Sigma$
 - Tensionless bilayer: $\Sigma = 0$
 - Leaflet tensions for binary mixture





• Leaflet tensions and flip-flops:



Miettinen, RL, Nanoletters (2019)

- •Add cholesterol
- Leaflet tensions relax towards $\Sigma_1 = \Sigma_2 = 0$

Leaflet Tensions and Nanovesicles

Ghosh, Satarifard, Grafmüller, RL, *Nanoletters* (2019) • Polymorphism of spherical nanovesicle:



- Four spherical vesicles
- Same volume
- Same total # of lipids
- Reduction of volume: very different shapes
- Shape transformations determined by leaflet tensions Σ₁ and Σ₂

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- R.

. . .

Endocytosis of nanoparticles by membranes Wetting and engulfment of droplets

Targeting Nanoparticles to Cells

• Nanoparticles (NPs) as drug delivery systems:



• Endocytic pathway also used by virusses, airborne ultrafine particles, ...



- Three steps: Adhesion, Complete Engulfment, Scission
- All steps governed by local stability relations

Agudo-Canalejo, RL: ACS Nano (2015); Soft Matter (2016)

39



- Attractive interactions between NP and membrane
- Van der Waals, electrostatic, receptor-ligand
- Gain of adhesion free energy but increase of elastic membrane energy
- Competition between adhesion and bending
- Bending rigidity κ versus adhesive strength *|W|*

Adhesion Length

- Adhesive strength |W| = adhesion free energy per area
- Bending rigidity κ and adhesive strength |W| define adhesion length $R_W = (2\kappa/|W|)^{1/2}$
- For specific NP-membane systems, R_W varies between 10 nm and 3 μ m !
- Large R_W values can be measured via membrane curvature along contact line



Onset of Adhesion: Key Parameters



- Three key parameters for onset of adhesion: Adhesion length R_W , Particle size R_{pa} , and Membrane curvature M at point of contact
- Membrane curvature *M* can be positive or negative:



Onset of Adhesion: Local Criterion

Agudo-Canalejo and RL, ACS Nano + Nano Letters (2015)

• Membrane starts to spread over particle if

$$M \le 1/R_{\rm W} - 1/R_{\rm pa} =: M_{\rm co}$$

contact curvature M_{co} is threshold value for M

• Example: $R_{\rm W} = R_{\rm pa}$ or $M_{\rm co} = 0$



• Large contact curvature M_{co} for small R_W or large |W|





- After onset of adhesion, membrane spreads over NP
- Membrane may engulf NP only partially or completely
- Complete engulfment involves closed membrane neck
- Necessary condition for complete engulfment: Closed membrane neck must be stable

Neck Stability: Local Criterion

Agudo-Canalejo and RL, ACS Nano + Nano Letters (2015)

• Closed membrane neck is stable if membrane curvature

$$M \ge 2m + 1/R_{\rm pa} - 1/R_{\rm W} =: M_{\rm ne}$$

2nd threshold value for *M*





Effective Constriction Force

$$M + M_{\rm co} - 2m + f (4\pi\kappa)^{-1} \ge 0$$

$$f_{\rm eff} = f + 4 \pi \kappa (M + M_{\rm co} - 2m) \ge 0$$

=> Engulfment force $f_{eng} = 100 \text{ pN}$

• Sufficient to create two hydrophobic bilayer edges



Receptor-Mediated Endocytosis

Chithrani et al, Nano Letters (2007)

- Uptake of gold nanoparticles by cells
- Particles bind to transferrin receptors
- Assembly of clathrin-coated vesicles Non-monotonic size-dependence !
 - Cell membrane with two types of segments, bound and unbound
 - Bound segment contains protein coat with spont curv $m_{bo} = -1/(40 \text{ nm})$
 - Good agreement with exp data: Agudo-Canalejo, RL: ACS Nano (2015)



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Biomolecular Condensates

Brangwynne ... Hyman, Science (2009)

- Biomolecular condensates that behave like liquid droplets
- Enriched in intrinsically disordered proteins (IDPs)
- Example for IDP: RNA-binding protein FUS
- Interaction of FUS-droplets with GUVs, two subsequent wetting transitions:



dewetting for high salt



partial wetting for intermediate salt



complete wetting for low salt ⁵¹

Two Wetting Transitions

• GUV + FUS-rich organelle + salt



Engulfment of Condensates

• Green FUS-rich condensates engulfed by red GUV:



Lipid Bilayer + Nanodroplet

Satarifard, Grafmüller, RL: ACS Nano (2018)

- Molecular simulations of lipid bilayer + nanodroplet
- Lateral box size L_{\parallel} determines mechanical tension
- Mechanical tension ~ size L_{\parallel} as control parameter



three aqueous phases α , β , γ α droplet (blue) coexists with β phase (white) three surface segments $\alpha\beta$, $\alpha\gamma$, $\beta\gamma$

Engulfment of Nanodroplets



• Tight-lipped membrane neck from negative line tension₅₅

Summary and Outlook

- Importance of spontaneous curvature
- Morphological complexity of GUVs uniform membranes => multi-domain membranes
- Controlled Division of GUVs binding of GFP => photoresponsive proteins
- GUVs a la carte:

dsGUVs with controlled spont curvature

- Controlling morphological complexity
- Smart storage and delivery systems



• Membranes, Theo

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